# The Reaction of Cyanoacetylhydrazine with $\omega$-Bromo(4methyl)acetophenone: Synthesis of Heterocyclic Derivatives with Antitumor Activity 

Rafat M. Mohareb ${ }^{1,2, *}$ and Abeer A. Mohamed ${ }^{3}$<br>1 Department of Organic Chemistry, Faculty of Pharmacy, October University for Modern Sciences and Arts, October City, A.R., Egypt<br>${ }^{2}$ Department of Chemistry, Faculty of Science, Cairo University, Giza, A.R., Egypt<br>${ }^{3}$ National Organization for Drug Control \& Research (NODCAR), P.O. 29, Cairo, Egypt; E-Mail: abeer_aam@yahoo.com (A.A.M.)

* Author to whom correspondence should be addressed; E-Mail: raafat_mohareb@yahoo.com; Tel: +202-37626269 or +202-35676570.

Received: 22 March 2010; in revised form: 14 April 2010 / Accepted: 21 April 2010 /
Published: 17 May 2010


#### Abstract

New approaches for the synthesis of hydrazide-hydrazone derivatives were demonstrated as well as some heterocyclizations of such derivatives to afford 1,3,4triazine, pyridine and 1,3,4-oxadiazine derivatives. The antitumor evaluation of the newly synthesized products against three cancer cell lines, namely breast adenocarcinoma (MCF-7), non-small cell lung cancer (NCI-H460) and CNS cancer (SF-268) were recorded. Most of the synthesized compounds showed high inhibitory effects.


Keywords: hydrazide-hydrazone; 1,3,4-triazine; pyridine; pyridazine; antitumor

## 1. Introduction

Hydrazide-hydrazones are an important class of compounds that has gained much importance in recent years due to their diverse biological activities [1-10]. The therapeutic prominence of hydrazidehydrazone derivatives is well established [11-12]. Hydrazide-hydrazones were also reported to elicit anticancer [13-20] and anti-HIV properties [21] and hence they have gained an important place in
medicinal chemistry. The discovery of the antineoplastic activity of the naturally occurring Schiff's bases has stimulated considerable research efforts in the field of condensed systems [22]. With the aim of constructing such condensed systems with the hydrazide-hydrazone nucleus, we turned our attention to using such compounds as synthons for heterocyclic derivatives and their anitumor evaluation [23,24].

## 2. Results and Discussion

In this work we report the reaction of cyanoacetylhydrazine (1) with $\omega$-bromo-(4-methylacetophenone) (2) in 1,4-dioxane which gave the condensed product 3 . The structure of compound 3 was confirmed based on analytical and spectral data. Thus, the ${ }^{1} \mathrm{H}$-NMR showed a singlet at $\delta 2.51$ for the $\mathrm{CH}_{3}$, two singlets at $\delta 4.31,4.72$ for the two $\mathrm{CH}_{2}$ groups, a multiplet at $\delta 6.50-7.76$ for the $\mathrm{C}_{6} \mathrm{H}_{4}$ group and a singlet at $\delta 11.46\left(\mathrm{D}_{2} \mathrm{O}\right.$ exchangeable) for the NH group. The reactivity of compound 3 towards different chemical reagents was studied. The reaction of $\mathbf{3}$ with either potassium cyanide or potassium thiocyanate gave the corresponding cyanide or the thiocyanate derivatives $\mathbf{4 a}$ and $\mathbf{4 b}$, respectively (Scheme 1).

Scheme 1. Synthesis of the hydrazide-hydrazone 3 and $\mathbf{4 a}, \mathbf{b}$.


The reaction of compound 3 with either hydrazine hydrate (5a) or phenylhydrazine (5b) gave the hydrazine derivative $\mathbf{6 a}$ or $\mathbf{6 b}$, respectively. Analytical and spectral data of the reaction products are in agreement with the proposed structures (see Experimental section). The reaction of either $\mathbf{6 a}$ or $\mathbf{6 b}$ with benzaldehyde (7) gave the benzal derivative 8a or 8b, respectively (Scheme 2).

On the other hand, the reaction of either $\mathbf{6 a}$ or $\mathbf{6 b}$ with either benzenediazonium chloride (9a), 3-cyano-4,5,6,7-tetrahydrobenzo[b]thiophene-2-diazonium chloride (9b) or ethyl 3-cyano-4,5,6,7-tetrahydrobenzo[b]thiophene-3-carboxylate-2-diazonium chloride (9c) gave the 3 -( $\alpha$-hydrazo-acetonitrilo)-1,2,4-triazine derivatives 10a-f, respectively (Scheme 3). The analytical and spectral data of the latter reaction products are all consistent with the proposed structures.

Scheme 2. Synthesis of the hydrazine derivatives $\mathbf{6 a}, \mathbf{b}$ and $\mathbf{8 a}, \mathbf{b}$.


Scheme 3. Synthesis of the 1,2,4-triazines 10a-f.


Next, we moved towards studying the reactivity of $\mathbf{6 a}$ and $\mathbf{6 b}$ with active methylene reagents. Thus, their reactions with either malononitrile (11a) or ethyl cyanoacetate (11b) gave the pyridine derivatives 12a-d, respectively (Scheme 4). The structures of the latter products were established on the basis of their analytical and spectral data. Thus, the ${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum of $\mathbf{1 2 c}$ showed a singlet at $\delta 2.51$ for the $\mathrm{CH}_{3}$ group, a singlet at $\delta 3.38$ for the $\mathrm{CH}_{2}$ group, two singlets at $\delta 4.38,4.79$ for the two $\mathrm{NH}_{2}$ groups and a multiplet at $\delta 6.48-8.19$ for the pyridine $\mathrm{H}-3, \mathrm{C}_{6} \mathrm{H}_{5}$ and $\mathrm{C}_{6} \mathrm{H}_{4}$, two singlets at $\delta 10.82,11.20$ for the two NH groups, respectively.

Scheme 4. Synthesis of 2-oxopyridines 12a-d.


| $\mathbf{1 2}$ | R | X |
| :---: | :---: | :---: |
| $\mathbf{a}$ | H | $\mathrm{NH}_{2}$ |
| $\mathbf{b}$ | H | OH |
| $\mathbf{c}$ | Ph | $\mathrm{NH}_{2}$ |
| $\mathbf{d}$ | Ph | OH |

On the other hand, the reaction of either $\mathbf{6 a}$ or $\mathbf{6} \mathbf{b}$ with either acetylacetone (13a) or ethyl acetoacetate (13b) gave the pyridine derivatives 14a-d, respectively (Scheme 5). The structures of the latter products were confirmed by their analytical and spectral data (see Experimental section).

Scheme 5. Synthesis of 2-oxopyridines 14a-d.


14a-d

| $\mathbf{1 4}$ | R | X |
| :---: | :---: | :--- |
| $\mathbf{a}$ | H | $\mathrm{CH}_{3}$ |
| $\mathbf{b}$ | H | OH |
| $\mathbf{c}$ | Ph | $\mathrm{CH}_{3}$ |
| $\mathbf{d}$ | Ph | OH |

Compound 3 underwent ready cyclization when heated in sodium ethoxide solution to give the 1,3,4-oxadiazine derivative 15 , whose structure was established from its analytical and spectral data (see Experimental section). The oxadiazine derivatives 15 seemed to be an intermediate for many reactions between 3 and many chemical reagents. Thus, the reaction of 3 with benzaldehyde (7) gave the 2-( $\alpha$-benzalacetonitrilo)-1,3,4-oxadiazine derivative 16. The analytical and spectral data of $\mathbf{1 6}$ were
in agreement with the proposed structure. Thus, the ${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum showed a singlet at $\delta 2.51$ for the $\mathrm{CH}_{3}$ group, a singlet at $\delta 4.22$ for the $\mathrm{CH}_{2}$ group, a singlet at $\delta 5.16$ for the $(=\mathrm{CH})$ group and a multiplet at $\delta 7.35-8.02$ for the $\mathrm{C}_{6} \mathrm{H}_{5}$ and $\mathrm{C}_{6} \mathrm{H}_{4}$ groups. The same product $\mathbf{1 6}$ was obtained through the reaction of compound 15 with benzaldehyde (7) (confirmed by m.p., mixed m.p. and fingerprint IR spectrum). On the other hand, the reaction of 15 with benzenediazonium chloride (9) gave the 2-( $\alpha-$ phenylhydrazo)-1,3,4-oxadiazine derivative 17 (Scheme 6).

Scheme 6. Synthesis of 1,3,4-oxadiazines 15, 16 and 17.

$3+\mathrm{PhCHO} \xrightarrow[\text { Piperidine }]{\text { Ethanol }}$
7


16
$15+\mathrm{PhCHO} \xrightarrow[\text { Piperidine }]{\text { Ethanol }}$
16

7


17

## Effect on the Growth of Human Tumor Cell Lines

The effect of compounds 4-17 was evaluated on the in vitro growth of three human tumor cell lines representing different tumor types, namely, breast adenocarcinoma (MCF-7), non-small cell lung cancer (NCI-H460) and CNS cancer (SF-268), after a continuous exposure of 48 h . The results are summarized in Table 1. All the compounds were able to inhibit the growth of the tested human tumor cell lines in a dose-dependent manner (data not shown). The results showed that compound $\mathbf{1 4 b}$, with its 2-hydroxypyridine group, showed the highest inhibitory effect against all the three tumor cell lines. In addition compounds $\mathbf{1 0 a}, \mathbf{1 0 f}, \mathbf{1 2 a}, \mathbf{1 2 b}$ and $\mathbf{1 4 c}$ showed high inhibitory effects. On the other hand, compounds $\mathbf{4 b}, \mathbf{6 a}, \mathbf{8 b}, \mathbf{1 0 b}, 10 d, 12 c, 12 d, 14 a, 14 c, 16$ and 17 showed the lowest inhibitory effect
towards adenocarcinoma (MCF-7). The rest of compounds showed the moderate growth inhibitory effects. Comparing compound $\mathbf{4 a}$ with $\mathbf{4} \mathbf{b}$, it is obvious that the presence of the $\alpha-S C N$ present in $\mathbf{4} \mathbf{b}$ showed lower inhibitory effect than $\mathbf{4 a}$ with it's $\alpha$-CN group. Comparing 1,2,4-triazine derivatives 10a (with the 4-phenylamino group) and $\mathbf{1 0 b}$ (with the 4-thiophenoamino group), the first has a greater inhibitory effect than the second towards the three cell lines.

Table 1. Effect of the newly synthesized products on the growth of three human tumor cell lines.

| Compound | $\mathbf{G I}_{\mathbf{5 0}}(\mu \mathrm{M})$ |  |  |
| :--- | :--- | :--- | :--- |
|  | MCF-7 | NCI-H460 | SF-268 |
| 4a | $10.0 \pm 0.6$ | $12.9 \pm 1.6$ | $15 \pm 1.1$ |
| 4b | $70.6 \pm 15.3$ | $38.1 \pm 10.8$ | $48.0 \pm 9.1$ |
| 6a | $50.8 \pm 18.5$ | $20.2 \pm 12.6$ | $50.0 \pm 8.7$ |
| 6b | $25.8 \pm 12.5$ | $22.4 \pm 8.6$ | $20.3 \pm 8.6$ |
| 8a | $24.6^{*}$ | $77.9 \pm 5.0$ | $35.0^{*}$ |
| 8b | $80.9 \pm 10.2$ | $70.9 \pm 6.1$ | $50.2 \pm 2.2$ |
| 10a | $8.0 \pm 0.6$ | $11.7 \pm 8.8$ | $14.8 \pm 1.6$ |
| 10b | $70.6 \pm 8.6$ | $55.6 \pm 5.8$ | $42.6 \pm 8.6$ |
| 10c | $24.2 \pm 0.6$ | $8.4 \pm 1.8$ | $10.4 \pm 8.4$ |
| 10d | $88.4 \pm 4.3$ | $66.8 \pm 12.0$ | $40.9 \pm 2.6$ |
| 10e | $12.4 \pm 2.6$ | $6.8 \pm 1.2$ | $8.3 \pm 9.2$ |
| 10f | $10.0 \pm 0.4$ | $14.9 \pm 1.6$ | $14 \pm 1.1$ |
| 12a | $12.4 \pm 1.6$ | $10.8 \pm 4.0$ | $16 \pm 6.0$ |
| 12b | $8.6 \pm 2.4$ | $10.4 \pm 6.2$ | $12.0 \pm 4.2$ |
| 12c | $86.8 \pm 6.0$ | $48.5 \pm 4.0$ | $38.4 \pm 2.6$ |
| 12d | $90.8 \pm 2.4$ | $78.2 \pm 2.2$ | $86.2 \pm 1.8$ |
| 14a | $66.4 \pm 6.0$ | $42.4 \pm 6.0$ | $62.3 \pm 6.0$ |
| 14b | $4.2 \pm 2.8$ | $10.2 \pm 6.8$ | $10.8 \pm 5.0$ |
| 14c | $80.3 \pm 4.0$ | $64.8 \pm 6.4$ | $50.8 \pm 6.4$ |
| 16 | $40.5 \pm 2.6$ | $32.6 \pm 4.0$ | $26.8 \pm 8.4$ |
| 17 | $60.4 \pm 6.0$ | $77.8 \pm 3.1$ | $47.0 \pm 6.4$ |

Results are given in concentrations that were able to cause $50 \%$ of cell growth inhibition $\left(\mathrm{GI}_{50}\right)$ after a continuous exposure of 48 h and show means $\pm$ SEM of three-independent experiments performed in duplicate. "Results from two-independent experiments performed in duplicate. Doxorubicin was used as positive control, $\mathrm{GI}_{50}$ : MCF-7 $=42.8 \pm 8.2 \mathrm{nM}$, NCI-H460 $=94.0 \pm 8.7 \mathrm{nM}$, and SF-268 $=94.0 \pm 7.0 \mathrm{nM}$.

## 3. Experimental

### 3.1. General

Melting points were determined on an Electrothermal melting point apparatus (Electrothermal 9100) and are uncorrected. IR spectra were recorded for KBr discs on a Pye Unicam SP-1000 spectrophotometer. ${ }^{1} \mathrm{H}-\mathrm{NMR} \&{ }^{13} \mathrm{C}$-NMR spectra were measured on a Varian EM-390-200 MHz in $\mathrm{CD}_{3} \mathrm{SOCD}_{3}$ as solvent using TMS as internal standard, and chemical shifts are expressed as $\delta$.

Analytical data were obtained from the Microanalytical Data Unit at Cairo University, Giza, Egypt. Antitumor evaluation for the newly synthesized products were performed by a research group at the National Research Center \& the National Cancer Institute at Cairo University.

4-Methyl- $\omega$-bromoacetophenone cyanoacetylhydrazone (3). To a solution of cyanoacetylhydrazine (1, $2.44 \mathrm{~g}, 0.02 \mathrm{~mol})$ in 1,4-dioxane ( 20 mL ), $\omega$-bromo-(4-methylacetophenone) $(5.24 \mathrm{~g}, 0.02 \mathrm{~mol})$ was added. The reaction mixture was stirred at room temperature for 1 hr then poured onto a beaker containing an ice/water mixture. The solid product formed was collected by filtration and dried obtaining pale yellow crystals (from ethanol).Yield: $5.02 \mathrm{~g}(71 \%)$, m.p. $148{ }^{\circ} \mathrm{C}$; IR ( KBr ) $\mathrm{v} / \mathrm{cm}^{-1}$ : 3400-3378 (NH), $3105\left(\mathrm{CH}\right.$ aromatic), $2956\left(\mathrm{CH}_{3}\right), 2259(\mathrm{CN}), 1681(\mathrm{C}=\mathrm{O}), 1610(\mathrm{C}=\mathrm{C}) ;{ }^{1} \mathrm{H}-\mathrm{NMR} \delta:$ $2.51\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 4.31,4.72\left(2 \mathrm{~s}, 4 \mathrm{H}, \mathrm{CH}_{2}\right), 6.50-7.76\left(\mathrm{~m}, 4 \mathrm{H}, \mathrm{C}_{6} \mathrm{H}_{4}\right), 11.46(\mathrm{~s}, 1 \mathrm{H}, \mathrm{NH}) . \mathrm{MS}:(\mathrm{m} / \mathrm{z})$ $293\left(\mathrm{M}^{+}\right)(6 \%), 200(28 \%), 117(100 \%), 68(69.9 \%) ;{ }^{13} \mathrm{C}-\mathrm{NMR}: 22.6\left(\mathrm{CH}_{3}\right), 27.0,58.7\left(2 \mathrm{CH}_{2}\right), 118.5$ (CN), 126.8, 128.4, 129.5, $138.8\left(\mathrm{C}_{6} \mathrm{H}_{5}\right), 156.3(\mathrm{C}=\mathrm{N})$, $173.8(\mathrm{C}=\mathrm{O})$; Anal. Calcd. for $\mathrm{C}_{12} \mathrm{H}_{12} \mathrm{~N}_{3} \mathrm{OBr}$ (294.24): C, 48.00; H, 4.11; N, 14.29\%. Found: C, 49.27; H, 4.45; N, 14.35\%.

### 3.2. General Procedure for the Synthesis of $\mathbf{4 a}$ or $\mathbf{4 b}$

To a solution of $3\left(0.54 \mathrm{~g}, 1.83 \times 10^{-3} \mathrm{~mol}\right)$ in ethanol $(25 \mathrm{~mL})$ in a water bath at $60^{\circ} \mathrm{C}$, either potassium cyanide $\left(0.11 \mathrm{~g}, 1.83 \times 10^{-3} \mathrm{~mol}\right)$ or potassium thiocyanate $\left(0.17 \mathrm{~g}, 1.83 \times 10^{-3} \mathrm{~mol}\right)$ was added with continuous stirring. The reaction mixture was left in the water bath for 30 min at $60^{\circ} \mathrm{C}$ then poured onto a beaker containing ice/water mixture and few drops of hydrochloric acid. The formed solid product was collected by filtration and dried.

4-Methyl- $\omega$-cyanoacetophenonecyanoacetylhydrazone (4a). Pale brown crystals (from ethanol). Yield: $0.20 \mathrm{~g}(62 \%)$, m.p. $160{ }^{\circ} \mathrm{C}$; IR ( KBr ) v/ $\mathrm{cm}^{-1}: 3450-3205(\mathrm{NH}), 3031(\mathrm{CH}$ aromatic), 2260, 2209 $(2 \mathrm{CN}), 1680(\mathrm{C}=\mathrm{O}) ;{ }^{1} \mathrm{H}$ NMR $\delta: 2.39\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 4.46,5.06\left(2 \mathrm{~s}, 4 \mathrm{H}, 2 \mathrm{CH}_{2}\right), 7.18-7.84(\mathrm{~m}, 4 \mathrm{H}$, $\left.\mathrm{C}_{6} \mathrm{H}_{4}\right), 10.82(\mathrm{~s}, 1 \mathrm{H}, \mathrm{NH}) ;{ }^{13} \mathrm{C}$ NMR: $23.8\left(\mathrm{CH}_{3}\right), 27.3,29.5\left(2 \mathrm{CH}_{2}\right), 116.9,117.0(2 \mathrm{CN}), 127.3,128.0$, 129.8, $138.9\left(\mathrm{C}_{6} \mathrm{H}_{4}\right), 155.8(\mathrm{C}=\mathrm{N}), 173.8(\mathrm{C}=\mathrm{O})$; Anal. Calcd. for $\mathrm{C}_{13} \mathrm{H}_{12} \mathrm{~N}_{4} \mathrm{O}$ (240.26): C, 64.99; H, 5.03 ; N, $23.32 \%$. Found: C, 64.79 ; H, 5.11 ; N, $22.99 \%$.

4-Methyl- $\omega$-thiocyanoacetophenoecyanoacetylhydrazone (4b). Buff crystals (from ethanol). Yield: $0.33 \mathrm{~g}(67 \%)$, m.p. $130{ }^{\circ} \mathrm{C}$; IR ( KBr ) v/cm ${ }^{-1}: 3445-3225(\mathrm{NH}), 3099\left(\mathrm{CH}\right.$ aromatic), $3034\left(\mathrm{CH}_{3}\right)$, 2969, $2925\left(2 \mathrm{CH}_{2}\right), 2264,2225(2 \mathrm{CN}), 1666(\mathrm{C}=\mathrm{O}), 1606(\mathrm{C}=\mathrm{C}) ;{ }^{1} \mathrm{H}-\mathrm{NMR} \delta: 2.49\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 4.50$, $5.08\left(2 \mathrm{~s}, 4 \mathrm{H}, 2 \mathrm{CH}_{2}\right), 7.06-7.93\left(\mathrm{~m}, 4 \mathrm{H}, \mathrm{C}_{6} \mathrm{H}_{4}\right), 11.52(\mathrm{~s}, 1 \mathrm{H}, \mathrm{NH}) ;{ }^{13} \mathrm{C}-\mathrm{NMR}: 23.8\left(\mathrm{CH}_{3}\right), 27.3,29.4$ $\left.\left(2 \mathrm{CH}_{2}\right), 116.7,117.8(2 \mathrm{CN}), 127.0,128.3,129.6,138.6\left(\mathrm{C}_{6} \mathrm{H}_{4}\right), 155.5 \mathrm{C}=\mathrm{N}\right), 173.9(\mathrm{C}=\mathrm{O})$; Anal. Calcd for $\mathrm{C}_{13} \mathrm{H}_{12} \mathrm{~N}_{4} \mathrm{OS}$ (272.32): C, 57.33; H, 4.44; N, 20.57; S, 11.77\%. Found: C, 57.38; H, 4.64; N, 20.68; S, 11.57\%.

### 3.3. General Procedure for the Synthesis of $\mathbf{6 a}$ and $\mathbf{6 b}$

To a solution of compound $3\left(1.50 \mathrm{~g}, 5.09 \times 10^{-3} \mathrm{~mol}\right)$ in ethanol ( 35 mL ) either hydrazine hydrate $\left(0.25 \mathrm{~g}, 5.09 \times 10^{-3} \mathrm{~mol}\right)$ or phenylhydrazine $\left(0.55 \mathrm{~g}, 5.09 \times 10^{-3} \mathrm{~mol}\right)$ was added. The reaction
mixture was heated under reflux for 3 hrs then poured onto a beaker containing an ice/water mixture and a few drops of hydrochloric acid. The formed solid product was collected by filtration and dried.

4-Methyl- $\omega$-hydrazinoacetophenonecyanoacetylhydrazone (6a). Greenish brown crystals (from ethanol). Yield: $0.65 \mathrm{~g}(52 \%)$, m.p. $120^{\circ} \mathrm{C}$; IR ( KBr ) v/cm ${ }^{-1}: 3400-3204\left(\mathrm{NH}_{2}, 2 \mathrm{NH}\right), 3027(\mathrm{CH}$ aromatic), $2918\left(\mathrm{CH}_{3}\right), 2203(\mathrm{CN}), 1688(\mathrm{C}=\mathrm{O}), 1607(\mathrm{C}=\mathrm{C}) ;{ }^{1} \mathrm{H}-\mathrm{NMR} \delta: 2.51\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 3.39$, $4.23\left(2 \mathrm{~s}, 4 \mathrm{H}, 2 \mathrm{CH}_{2}\right), 4.82\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{NH}_{2}\right), 6.87-8.40\left(\mathrm{~m}, 4 \mathrm{H}, \mathrm{C}_{6} \mathrm{H}_{4}\right), 9.26,11.41(2 \mathrm{~s}, 2 \mathrm{H}, 2 \mathrm{NH})$; ${ }^{13} \mathrm{C}-\mathrm{NMR}: 23.6\left(\mathrm{CH}_{3}\right), 27.0,29.6\left(2 \mathrm{CH}_{2}\right), 127.8,128.3,129.5,138.6\left(\mathrm{C}_{6} \mathrm{H}_{4}\right), 155.6,164.2(2 \mathrm{C}=\mathrm{N})$, 173.8 (C=O); Anal. Calcd. for $\mathrm{C}_{12} \mathrm{H}_{15} \mathrm{~N}_{5} \mathrm{O}$ (245.28): C, 58.76 ; H, 6.16; N, 28.55\%. Found: C, 58.55; H, 5.96; N, 28.32\%.

4-Methyl- $\omega$-phenylhydrazinoacetophenonecyanoacetylhydrazone (6b). Brown crystals (from ethanol). Yield: $0.95 \mathrm{~g}(63 \%)$, m.p. $72{ }^{\circ} \mathrm{C}$; IR ( KBr ) v/cm ${ }^{-1}: 3500-3174$ (3NH), 3026 ( CH aromatic), 2969 $\left(\mathrm{CH}_{3}\right), 2205(\mathrm{CN}), 1687(\mathrm{C}=\mathrm{O}), 1600(\mathrm{C}=\mathrm{C}) ;{ }^{1} \mathrm{H}-\mathrm{NMR} \delta: 2.51\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 3.39,4.35\left(2 \mathrm{~s}, 4 \mathrm{H}, 2 \mathrm{CH}_{2}\right)$, 6.79-8.40 (m, $\left.9 \mathrm{H}, \mathrm{C}_{6} \mathrm{H}_{5}, \mathrm{C}_{6} \mathrm{H}_{4}\right), 10.82,11.19,12.63(3 \mathrm{~s}, 3 \mathrm{H}, 3 \mathrm{NH}) ;{ }^{13} \mathrm{C}-\mathrm{NMR}: 23.6\left(\mathrm{CH}_{3}\right), 27.0,29.9$ $\left(2 \mathrm{CH}_{2}\right), 120.3,122.5,127.3,128.4,129.3,138.5,140.1\left(\mathrm{C}_{6} \mathrm{H}_{4}, \mathrm{C}_{6} \mathrm{H}_{5}\right), 155.6,164.2(2 \mathrm{C}=\mathrm{N}), 173.8$ (C=O); Anal. Calcd. for $\mathrm{C}_{18} \mathrm{H}_{19} \mathrm{~N}_{5} \mathrm{O}$ (321.38): C, 67.27; H, 5.96; N, 21.79\%. Found: C, 67.37; H, 5.84; N, 22.61\%.
$\alpha$-Benzal-4-methyl- $\omega$-hydrazinoacetophenonecyanoacetylhydrazone (8a): To a solution of compound $6 \mathbf{a}\left(0.29 \mathrm{~g}, 1.18 \times 10^{-3} \mathrm{~mol}\right)$ in ethanol $(25 \mathrm{~mL})$ containing piperidine $(0.5 \mathrm{~mL})$, benzaldehyde $(0.11 \mathrm{~g}$, $1.48 \times 10^{-3} \mathrm{~mol}$ ) was added. The reaction mixture was heated under reflux for 3 hrs then poured onto a beaker containing an ice/water mixture and few drops of hydrochloric acid. The formed solid product was collected by filtration and dried obtaining deep yellowish brown crystals (from ethanol).Yield: $0.26 \mathrm{~g}(66 \%)$, m.p. $90^{\circ} \mathrm{C}$; IR ( KBr ) v/cm ${ }^{-1}: 3500-3194\left(\mathrm{NH}_{2}, 2 \mathrm{NH}\right), 3027\left(\mathrm{CH}\right.$ aromatic), $2919\left(\mathrm{CH}_{3}\right)$, $2218(\mathrm{CN}), 1680(\mathrm{C}=\mathrm{O}), 1608(\mathrm{C}=\mathrm{C}) ;{ }^{1} \mathrm{H}-\mathrm{NMR} \delta: 2.51\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 3.36\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 4.77(\mathrm{~s}, 2 \mathrm{H}$, $\left.\mathrm{NH}_{2}\right), 5.99(\mathrm{~s}, 1 \mathrm{H},=\mathrm{CH}), 7.27-8.40\left(\mathrm{~m}, 9 \mathrm{H}, \mathrm{C}_{6} \mathrm{H}_{5}, \mathrm{C}_{6} \mathrm{H}_{4}\right), 8.81,9.15(2 \mathrm{~s}, 2 \mathrm{H}, 2 \mathrm{NH}) ;{ }^{13} \mathrm{C}-\mathrm{NMR}: 23.2$ $\left(\mathrm{CH}_{3}\right), 27.4,29.2\left(2 \mathrm{CH}_{2}\right), 122.3,122.8,122.5,126.0128 .3,129.9,138.3\left(\mathrm{C}_{6} \mathrm{H}_{5}, \mathrm{C}_{6} \mathrm{H}_{4}\right) 155.0,164.3$ $(2 \mathrm{C}=\mathrm{N}), 174.3(\mathrm{C}=\mathrm{O})$; Anal. Calcd. for $\mathrm{C}_{19} \mathrm{H}_{19} \mathrm{~N}_{5} \mathrm{O}$ (333.39): C, 68.45 ; H, $5.74 ; \mathrm{N}, 21.01 \%$. Found: C, 68.55; H, 5.54; N, 21.31\%.
$\alpha$-Benzal-4-methyl- $\omega$-phenylhydrazinoacetophenonecyanoacetyl-hydrazone (8b). To a solution of compound $6 \mathbf{b}\left(0.36 \mathrm{~g}, 1.12 \times 10^{-3} \mathrm{~mol}\right)$ in ethanol $(25 \mathrm{~mL})$ containing piperidine $(0.5 \mathrm{~mL})$, benzaldehyde $\left(0.11 \mathrm{~g}, 1.12 \times 10^{-3} \mathrm{~mol}\right)$ was added. The reaction mixture was heated under reflux for 3 hrs then poured onto a beaker containing an ice/water mixture and a few drops of hydrochloric acid. The formed solid product was collected by filtration and dried obtaining yellowish brown crystals (from ethanol). Yield: $0.33 \mathrm{~g}(73 \%)$, m.p. $102{ }^{\circ} \mathrm{C}$; IR ( KBr ) v/cm ${ }^{-1}: 3300-3179(3 \mathrm{NH}), 3026(\mathrm{CH}$ aromatic), $2920\left(\mathrm{CH}_{3}\right), 2210(\mathrm{CN}), 1688(\mathrm{C}=\mathrm{O}), 1600(\mathrm{C}=\mathrm{C}) ;{ }^{1} \mathrm{H}$ NMR $\delta: 2.88\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 3.37(\mathrm{~s}$, $\left.2 \mathrm{H}, \mathrm{CH}_{2}\right), 6.90-8.40\left(\mathrm{~m}, 15 \mathrm{H},=\mathrm{CH}, 2 \mathrm{C}_{6} \mathrm{H}_{5}, \mathrm{C}_{6} \mathrm{H}_{4}\right), 9.20,10.83,12.62(3 \mathrm{~s}, 3 \mathrm{H}, 3 \mathrm{NH}) ;{ }^{13} \mathrm{C}-\mathrm{NMR}: 23.2$ $\left(\mathrm{CH}_{3}\right), 27.4,29.2\left(2 \mathrm{CH}_{2}\right), 120.9,121.1,122.3,122.8,123.4,123.9,122.5,126.0128 .3,129.5,138.6$ $\left(2 \mathrm{C}_{6} \mathrm{H}_{5}, \mathrm{C}_{6} \mathrm{H}_{4}\right) 155.3,164.0(2 \mathrm{C}=\mathrm{N})$, $174.6(\mathrm{C}=\mathrm{O})$; Anal. Calcd. for $\mathrm{C}_{25} \mathrm{H}_{23} \mathrm{~N}_{5} \mathrm{O}$ (409.48): C, 73.33; H, 5.66 ; N, $17.10 \%$. Found: C, 73.52; H, 5.97; N, 17.33\%.

1-Amino-6-(4-methylphenyl)-3-( $\alpha$-phenylhydrazoacetonitrilo)-1,2,4-triazine (10a). To a cold solution $\left(0-5^{\circ} \mathrm{C}\right)$ of compound $\mathbf{6 a}\left(0.40 \mathrm{~g}, 1.63 \times 10^{-3} \mathrm{~mol}\right)$ in ethanol $(50 \mathrm{~mL})$ containing sodium hydroxide $(10 \mathrm{~mL}, 10 \%)$ and a solution of benzenediazonium chloride $\left(1.63 \times 10^{-3} \mathrm{~mol}\right)$ [which was prepared by dissolving sodium nitrite $\left(0.16 \mathrm{~g}, 2.44 \times 10^{-3} \mathrm{~mol}\right)$ in water, 2 mL was added to a cold solution of aniline $\left(0.15 \mathrm{~g}, 1.63 \times 10^{-3} \mathrm{~mol}\right)$ containing the appropriate amount of hydrochloric acid and with continuous stirring] was added with continuous stirring. The formed solid product was collected by filtration and dried to give reddish brown crystals (from ethanol and few drops of dimethylformamide). Yield: $0.58 \mathrm{~g}(63 \%)$, m.p. $170{ }^{\circ} \mathrm{C}$; IR ( KBr ) v/cm ${ }^{-1}: 3400-3307\left(\mathrm{NH}_{2}, \mathrm{NH}\right), 3027(\mathrm{CH}$ aromatic), 2919, $2862\left(\mathrm{CH}_{3}, \mathrm{CH}_{2}\right), 2211(\mathrm{CN}), 1600(\mathrm{C}=\mathrm{C}) ;{ }^{1} \mathrm{H}-\mathrm{NMR} \delta: 2.51\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 3.37\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}\right)$, $4.22\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{NH}_{2}\right), 6.50-8.20\left(\mathrm{~m}, 9 \mathrm{H}, \mathrm{C}_{6} \mathrm{H}_{5}, \mathrm{C}_{6} \mathrm{H}_{4}\right), 11.64(\mathrm{~s}, 1 \mathrm{H}, \mathrm{NH}) ;{ }^{13} \mathrm{C}-\mathrm{NMR}: 23.8\left(\mathrm{CH}_{3}\right), 51.9$ (triazine $\mathrm{CH}_{2}$ ), $115.8(\mathrm{CN}), 118.3,118.9,119.2,121.6,124.8,133.0,139.6\left(\mathrm{C}_{6} \mathrm{H}_{5}, \mathrm{C}_{6} \mathrm{H}_{4}\right), 156.9,163.8$ 164.1 ( $3 \mathrm{C}=\mathrm{N}$ ); Anal. Calcd. for $\mathrm{C}_{18} \mathrm{H}_{17} \mathrm{~N}_{7}$ (331.38): C, 65.24 ; H, 5.17; N, 29.58\%. Found: C, 65.42; H, 5.27; N, 29.36\%.

### 3.4. General Procedure for the Synthesis of 10b and 10c

To a cold solution $\left(0-5{ }^{\circ} \mathrm{C}\right)$ of compound $\mathbf{6 a}\left(0.49 \mathrm{~g}, 1.99 \times 10^{-3} \mathrm{~mol}\right)$ in ethanol ( 50 mL ) containing sodium hydroxide solution ( $10 \mathrm{~mL}, 10 \%$ ) was added with continuous stirring a solution of either 3-cyano-4,5,6,7-terahydrobenzo[b]thiophene-2-diazonium chloride ( $\mathbf{9 b}$ ) $\left(1.99 \times 10^{-3} \mathrm{~mol}\right)$ or ethyl 4,5,6,7-tetrahydrobenzo[b]thiophene-3-carboxylate-2-diazonium chloride (9c) $\left(1.99 \times 10^{-3} \mathrm{~mol}\right)$ [which was prepared by dissolving sodium nitrite $\left(0.20 \mathrm{~g}, 2.99 \times 10^{-3} \mathrm{~mol}\right)$ in water, 2 mL was added to a cold solution of either the 2-amino-3-cyano-4,5,6,7-tetrahydrobenzo[b]thiophene $(0.35 \mathrm{~g}$, $1.99 \times 10^{-3} \mathrm{~mol}$ ) or ethyl 2-amino-4,5,6,7-tetrahydrobenzo[b]thiophene-3-carboxylate ( 0.45 g , $1.99 \times 10^{-3} \mathrm{~mol}$ ) dissolved in acetic acid ( 50 mL ) containing the appropriate amount of hydrochloric acid and with continuous stirring]. The solid product formed was collected by filtration and dried.

1-Amino-6-(4-methylphenyl)-3-( $\alpha-(3-c y a n o-2-h y d r a z o-4,5,6,7-t e t r a-h y d r o b e n z o[b] t h i o p h e n e) a c e t o-~$ nitrilo)-1,2,4-triazine (10b). Deep brown crystals (from ethanol). Yield: $0.67 \mathrm{~g}(75 \%)$, m.p. $210-214{ }^{\circ} \mathrm{C}$; IR ( KBr ) v/cm ${ }^{-1}: 3600-3425\left(\mathrm{NH}_{2}, \mathrm{NH}\right), 3030\left(\mathrm{CH}\right.$ aromatic), $2930\left(\mathrm{CH}_{3}\right), 2250,2217(2 \mathrm{CN}), 1606$ $(\mathrm{C}=\mathrm{C}) ;{ }^{1} \mathrm{H}-\mathrm{NMR} \delta: 1.63-2.34\left(\mathrm{~m}, 8 \mathrm{H}\right.$, cyclohexene $\left.4 \mathrm{CH}_{2}\right), 2.51\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 3.60\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 4.78$ (s, $2 \mathrm{H}, \mathrm{NH}_{2}$ ), 6.97-8.32 (m, 4H, $\left.\mathrm{C}_{6} \mathrm{H}_{4}\right), 9.84(\mathrm{~s}, 1 \mathrm{H}, \mathrm{NH}) ;{ }^{13} \mathrm{C}-\mathrm{NMR}: 19.4,23.0,25.1,27.8$ (cyclohexene $4 \mathrm{CH}_{2}$ ), $24.8\left(\mathrm{CH}_{3}\right), 51.9$ (triazine $\left.\mathrm{CH}_{2}\right), 115.8,116.4(2 \mathrm{CN}), 118.3,118.9,119.2,121.6$, 124.6, 133.2, 136.5, 136.9139 .6 (thiophene C, $\mathrm{C}_{6} \mathrm{H}_{5}, \mathrm{C}_{6} \mathrm{H}_{4}$ ), 156.0, 163.5164 .3 (3 C=N); Anal. Calcd. for $\mathrm{C}_{21} \mathrm{H}_{20} \mathrm{~N}_{8} \mathrm{~S}$ (416.50): C, 60.56 ; H, 4.84; N, 26.90; S, 7.70\%. Found: C, $60.28 ; \mathrm{H}, 5.03 ; \mathrm{N}, 26.61$; S, 7.88\%.

Ethyl 1-amino-6-(4-methylphenyl)-3-( $\alpha$ (2-hydrazo-4,5,6,7-tetra-hydrobenzo[b]thiophene-3-carboxyl-ate)acetonitrilo)-1,2,4-triazine (10c). Brown crystals (from ethanol) Yield: $0.53 \mathrm{~g}(57 \%)$, m.p. $90^{\circ} \mathrm{C}$; IR ( KBr ) v/cm ${ }^{-1}$ : 3400-3287 ( $\left.\mathrm{NH}_{2}, \mathrm{NH}\right), 2976,2934,2860\left(2 \mathrm{CH}_{3}, \mathrm{CH}_{2}\right), 2213(\mathrm{CN}), 1711(\mathrm{C}=\mathrm{O})$; ${ }^{1} \mathrm{H}-\mathrm{NMR}$ : $1.07-1.92\left(\mathrm{~m}, 8 \mathrm{H}\right.$, cyclohexene $\left.4 \mathrm{CH}_{2}\right), 2.51\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 2.77\left(\mathrm{t}, 3 \mathrm{H}, J=7.04 \mathrm{~Hz}, \mathrm{CH}_{3}\right)$, $3.36\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 4.24\left(\mathrm{q}, 2 \mathrm{H}, J=7.04 \mathrm{~Hz}, \mathrm{CH}_{2}\right), 4.99\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{NH}_{2}\right), 6.50-8.08\left(\mathrm{~m}, 4 \mathrm{H}, \mathrm{C}_{6} \mathrm{H}_{4}\right), 9.88$ (s, 1H, NH); ${ }^{13} \mathrm{C}$-NMR: 14.5, $24.6\left(2 \mathrm{CH}_{3}\right.$ ), 23.2, 23.6, 25.1, 27.6 (cyclohexene $4 \mathrm{CH}_{2}$ ), 55.4 (triazine
$\left.\mathrm{CH}_{2}\right), 60.8$ (ester $\left.\mathrm{CH}_{2}\right), 128.7,129.0,129.7,132.1,136.5,138.9,140.8\left(\mathrm{C}_{6} \mathrm{H}_{5}\right.$, thiophene C), 115.9, 116.7 (2CN), 155.2, 158.0, 163.1 ( $3 \mathrm{C}=\mathrm{N}$ ), 165.6 (CO); Anal. Calcd. for $\mathrm{C}_{23} \mathrm{H}_{25} \mathrm{~N}_{7} \mathrm{O}_{2} \mathrm{~S}$ (463.56): C, 59.59; H, 5.44; N, 21.15; S, 6.93\%. Found: C, 59.83; H, 5.46; N, 20.91; S, 7.05\%.

1-Phenylamino-6-(4-methylphenyl)-3-( $\alpha-$ phenylhydrazoacetonitrilo)-1,2,4-triazine (10d). To a cold solution $\left(0-5^{\circ} \mathrm{C}\right)$ of compound $\mathbf{6 b}\left(0.53 \mathrm{~g}, 1.64 \times 10^{-3} \mathrm{~mol}\right)$ in ethanol $(50 \mathrm{~mL})$ containing sodium hydroxide ( $10 \mathrm{~mL}, 10 \%$ ) was added with continuous stirring a solution of benzenediazonium chloride $\left(1.64 \times 10^{-3} \mathrm{~mol}\right)$ [which was prepared by dissolving sodium nitrite $\left(0.17 \mathrm{~g}, 2.47 \times 10^{-3} \mathrm{~mol}\right)$ in water, 2 mL was added to a cold solution of aniline ( $0.15 \mathrm{~g}, 1.64 \times 10^{-3} \mathrm{~mol}$ ) containing the appropriate amount of hydrochloric acid and with continuous stirring]. The formed solid product was collected by filtration to give reddish brown crystals (from ethanol and few drops of dimethylformamide). Yield: $0.58 \mathrm{~g}(86 \%)$, m.p. $120^{\circ} \mathrm{C}$; IR ( KBr ) $\mathrm{v} / \mathrm{cm}^{-1}: 3500-3422(2 \mathrm{NH}), 3057(\mathrm{CH}$ aromatic), 3028, $2921\left(\mathrm{CH}_{3}, \mathrm{CH}_{2}\right), 2211(\mathrm{CN}), 1600(\mathrm{C}=\mathrm{C}) ;{ }^{1} \mathrm{H}-\mathrm{NMR} \delta: 2.50\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 3.31\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 6.60-7.65$ (m, 14H, $\left.2 \mathrm{C}_{6} \mathrm{H}_{5}, \mathrm{C}_{6} \mathrm{H}_{4}\right), 8.16,9.05(\mathrm{~s}, 2 \mathrm{H}, 2 \mathrm{NH}) ;{ }^{13} \mathrm{C}-\mathrm{NMR}: 23.6\left(\mathrm{CH}_{3}\right), 51.9$ (triazine $\left.\mathrm{CH}_{2}\right), 115.6$ (CN), 118.6, 118.7, 119.2, 120.1, 121.6, 124.8, 133.6, 134.8, $139.9\left(2 \mathrm{C}_{6} \mathrm{H}_{5}, \mathrm{C}_{6} \mathrm{H}_{4}\right), 156.6,163.7,164.0$ (3 C=N); Anal. Calcd. for $\mathrm{C}_{24} \mathrm{H}_{21} \mathrm{~N}_{7}$ (407.47): C, $70.74 ; \mathrm{H}, 5.19 ; \mathrm{N}, 24.06 \%$. Found: C, 71.05; H, 5.38; N, 23.87\%.

### 3.5. General Procedure for the Synthesis of 10e and $\mathbf{1 0 f}$

To a cold solution $\left(0-5{ }^{\circ} \mathrm{C}\right)$ of compound $\mathbf{6 b}\left(0.40 \mathrm{~g}, 1.24 \times 10^{-3} \mathrm{~mol}\right)$ in ethanol ( 50 mL ) containing sodium hydroxide solution ( $10 \mathrm{~mL}, 10 \%$ ) was added with continuous stirring a solution of either 3-cyano-4,5,6,7-terahydrobenzo[b]thiophene-2-diazonium chloride (9b) (1.24 $\left.\times 10^{-3} \mathrm{~mol}\right)$ or ethyl 4,5,6,7-tetrahydrobenzo[b]thiophen-3-carboxylate-2-diazonium chloride (9c) $\left(1.24 \times 10^{-3} \mathrm{~mol}\right)$ [which was prepared by dissolving sodium nitrite $\left(0.12 \mathrm{~g}, 1.86 \times 10^{-3} \mathrm{~mol}\right)$ in water, 2 mL was added to a cold solution of either the 2-amino-3-cyano-4,5,6,7-tetrahydrobenzo[b]thiophene $(0.22 \mathrm{~g}$, $1.99 \times 10^{-3} \mathrm{~mol}$ ) or ethyl 2-amino-4,5,6,7-tetrahydrobenzo[b]thiophene-3-carboxylate ( 0.28 g , $1.99 \times 10^{-3} \mathrm{~mol}$ ) dissolved in acetic acid ( 50 mL ) containing the appropriate amount of hydrochloric acid and with continuous stirring]. The solid product formed was collected by filtration and dried.

## 1-Phenylamino-6-(4-methylphenyl)-3-( $\alpha-(3-c y a n o-2-h y d r a z o-4,5,6,7-t e t r a h y d r o b e n z o[b] t h i o p h e n e)-~$

acetonitrilo)-1,2,4-triazine (10e). Reddish brown crystals (from ethanol), yield: $0.34 \mathrm{~g}(55 \%)$, m.p. $190{ }^{\circ} \mathrm{C}$; IR ( KBr ) v/cm ${ }^{-1}: 3500-3425(2 \mathrm{NH}), 3030\left(\mathrm{CH}\right.$ aromatic), $2930\left(\mathrm{CH}_{3}\right), 2250,2217(2 \mathrm{CN}), 1606$ ( $\mathrm{C}=\mathrm{C}$ ) ; ${ }^{1} \mathrm{H}-\mathrm{NMR} \delta: 1.63-2.34\left(\mathrm{~m}, 8 \mathrm{H}\right.$, cyclohexene), $2.51\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 3.60\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 6.97-8.32$ (m, 9H, $\left.\mathrm{C}_{6} \mathrm{H}_{5}, \mathrm{C}_{6} \mathrm{H}_{4}\right), 9.84,10.00(\mathrm{~s}, 2 \mathrm{H}, 2 \mathrm{NH}) ;{ }^{13} \mathrm{C}-\mathrm{NMR}: 19.1,23.3,25.4,27.4$ (cyclohexene $4 \mathrm{CH}_{2}$ ), $24.9\left(\mathrm{CH}_{3}\right), 51.7\left(\right.$ triazine $\left.\mathrm{CH}_{2}\right), 115.6,116.3(2 \mathrm{CN}), 118.3,118.9,119.2,121.6,124.6,133.2,136.5$, 136.9139 .4 (thiophene $\mathrm{C}, \mathrm{C}_{6} \mathrm{H}_{5}, \mathrm{C}_{6} \mathrm{H}_{4}$ ), 156.1, 163. 3, 164.6 ( $3 \mathrm{C}=\mathrm{N}$ ). Anal. Calcd. for $\mathrm{C}_{27} \mathrm{H}_{24} \mathrm{~N}_{8} \mathrm{~S}$ (492.60): C, $65.83 ;$ H, 4.91 ; N, 22.75; S, $6.51 \%$. Found: C, 65.96 ; H, $5.24 ;$ N, $22.55 ;$ S, $6.79 \%$.

Ethyl 1-phenylamino-6-(4-methylphenyl)-3-(a(2-hydrazo-4,5,6,7-tetrahydrobenzo-[b]thiophen-3-carboxylate)acetonitrilo)-1,2,4-triazine (10f). Pale reddish brown crystals (from ethanol). Yield: $0.40 \mathrm{~g}(59 \%)$, m.p. $150-160^{\circ} \mathrm{C}$; IR (KBr) v/cm ${ }^{-1}: 3400-3287(2 \mathrm{NH}), 2976,2934,2860\left(2 \mathrm{CH}_{3}\right.$,
$\mathrm{CH}_{2}$ ), $2213(\mathrm{CN}), 1711(\mathrm{C}=\mathrm{O})$; ${ }^{1} \mathrm{H}-\mathrm{NMR} \delta: 1.07-1.92\left(\mathrm{~m}, 8 \mathrm{H}\right.$, cyclohexene), $2.51\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 2.77$ (t, $3 \mathrm{H}, J=6.89 \mathrm{~Hz}, \mathrm{CH}_{3}$ ), $3.36\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 4.24\left(\mathrm{q}, 2 \mathrm{H}, J=6.89 \mathrm{~Hz}, \mathrm{CH}_{2}\right), 6.50-8.08\left(\mathrm{~m}, 9 \mathrm{H}, \mathrm{C}_{6} \mathrm{H}_{5}\right.$, $\left.\mathrm{C}_{6} \mathrm{H}_{4}\right), 9.88,10.30(2 \mathrm{~s}, 2 \mathrm{H}, 2 \mathrm{NH})$; ${ }^{13} \mathrm{C}-\mathrm{NMR}: 14.3,24.8\left(2 \mathrm{CH}_{3}\right), 23.0,23.3,25.1,27.8$ (cyclohexene $4 \mathrm{CH}_{2}$ ), 55.54 (triazine $\mathrm{CH}_{2}$ ), 60.8 (ester $\mathrm{CH}_{2}$ ), 128.7, 129.2, 129.7, 132.0, 136.5, 138.9, $140.8\left(\mathrm{C}_{6} \mathrm{H}_{5}\right.$, thiophene C), 115.8, $116.5(2 \mathrm{CN}), 155.2,158.0,163.3(3 \mathrm{C}=\mathrm{N}), 165.6(\mathrm{CO})$; Anal. Calcd. for $\mathrm{C}_{29} \mathrm{H}_{29} \mathrm{~N}_{7} \mathrm{O}_{2} \mathrm{~S}$ (539.65): C, 64.54; H, 5.42; N, 18.17; S, 5.94\%. Found: C, 64.68; H, 5.23; N, 17.97; S, $6.20 \%$.

### 3.6. General Procedure for the Synthesis of 12a and 12b

To a solution of compound $\mathbf{6 a}\left(0.47 \mathrm{~g}, 1.91 \times 10^{-3} \mathrm{~mol}\right)$ in ethanol ( 20 mL ) containing triethylamine $(0.5 \mathrm{~mL})$, either malononitrile $\left(0.12 \mathrm{~g}, 1.91 \times 10^{-3} \mathrm{~mol}\right)$ or ethyl cyanoacetate $\left(0.21 \mathrm{~g}, 1.91 \times 10^{-3} \mathrm{~mol}\right)$ was added. The reaction mixture was heated under reflux for 3 hrs then poured onto a beaker containing an ice/water mixture and a few drops of hydrochloric acid. The solid product formed was collected by filtration and dried.

3-Cyano-4,6-diamino-2-oxo-1-imino-(4-methyl- $\omega$-hydrazinoaceto-phenonylidieno)pyridine (12a). Brown crystals (from ethanol). Yield: $0.38 \mathrm{~g}(64 \%)$, m.p. $158^{\circ} \mathrm{C}$; IR ( KBr ) v/cm ${ }^{-1}: 3500-3227\left(3 \mathrm{NH}_{2}\right.$, NH), $3028\left(\mathrm{CH}\right.$ aromatic), $2974\left(\mathrm{CH}_{3}\right), 2919\left(\mathrm{CH}_{2}\right), 2201(\mathrm{CN}), 1685(\mathrm{C}=\mathrm{O}), 1600(\mathrm{C}=\mathrm{C}) ;{ }^{1} \mathrm{H}-\mathrm{NMR} \delta$ : $2.51\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 3.38\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 4.38,4.79,5.30\left(3 \mathrm{~s}, 6 \mathrm{H}, 3 \mathrm{NH}_{2}\right), 7.05-8.19(\mathrm{~m}, 5 \mathrm{H}$, pyridine $\mathrm{H}-3$, $\left.\mathrm{C}_{6} \mathrm{H}_{4}\right), 10.82(\mathrm{~s}, 1 \mathrm{H}, \mathrm{NH}) ;{ }^{13} \mathrm{C}-\mathrm{NMR}: ~ 23.8\left(\mathrm{CH}_{3}\right), 51.7\left(\mathrm{CH}_{2}\right), 80.9,89.5,114.8,125.6$ (pyridine C), $117.9(\mathrm{CN}), 126.2,128.0,129.3,129.6\left(\mathrm{C}_{6} \mathrm{H}_{5}\right), 163.5(\mathrm{C}=\mathrm{O}), 170.9(\mathrm{C}=\mathrm{N})$; Anal. Calcd. for $\mathrm{C}_{15} \mathrm{H}_{17} \mathrm{~N}_{7} \mathrm{O}$ (311.34): C, 57.87 ; H, 5.50; N, 31.49\%. Found: C, 57.96 ; H, 5.48 ; N, 31.68\%.

## 4-Amino-3-cyano-6-hydroxy-2-oxo-1-imino(4-methyl- $\omega$-hydrazino-acetophenonylidieno)pyridine

(12b). Pale brown crystals (from ethanol).Yield: $0.66 \mathrm{~g}(111 \%)$, m.p. $140{ }^{\circ} \mathrm{C}$; IR ( KBr ) $\mathrm{v} / \mathrm{cm}^{-1}$ : $3600-3227\left(\mathrm{OH}, 2 \mathrm{NH}_{2}, \mathrm{NH}\right), 3028\left(\mathrm{CH}\right.$ aromatic), $2974\left(\mathrm{CH}_{3}\right), 2919\left(\mathrm{CH}_{2}\right), 2201(\mathrm{CN}), 1685(\mathrm{C}=\mathrm{O})$, $1600(\mathrm{C}=\mathrm{C}) ;{ }^{1} \mathrm{H}-\mathrm{NMR} \delta: 2.50\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 3.39\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 4.34,4.77,\left(2 \mathrm{~s}, 4 \mathrm{H}, 2 \mathrm{NH}_{2}\right), 7.05-8.19$ $\left(\mathrm{m}, 5 \mathrm{H}\right.$, pyridine $\left.\mathrm{H}-3, \mathrm{C}_{6} \mathrm{H}_{4}\right), 10.84(\mathrm{~s}, 1 \mathrm{H}, \mathrm{NH}), 12.64(\mathrm{~s}, 1 \mathrm{H}, \mathrm{OH}) ;{ }^{13} \mathrm{C}-\mathrm{NMR}: 23.6\left(\mathrm{CH}_{3}\right), 51.4\left(\mathrm{CH}_{2}\right)$, $82.9,89.5,116.8,125.9$ (pyridine C), $117.9(\mathrm{CN}), 126.4,128.0,129.3,129.6\left(\mathrm{C}_{6} \mathrm{H}_{5}\right), 170.9,176.5$ $(2 \mathrm{C}=\mathrm{N})$, $164.3(2 \mathrm{C}=\mathrm{O})$; Anal. Calcd. for $\mathrm{C}_{15} \mathrm{H}_{16} \mathrm{~N}_{6} \mathrm{O}_{2}$ (312.33): C, $57.68 ; \mathrm{H}, 5.16 ; \mathrm{N}, 26.90 \%$. Found: C, 57.42; H, 5.09; N, 26.76\%.

### 3.7. General Procedure for the Synthesis of 12c and 12d

To a solution of compound $\mathbf{6 b}\left(0.60 \mathrm{~g}, 1.86 \times 10^{-3} \mathrm{~mol}\right)$ in ethanol ( 20 mL ) containing triethylamine $(0.5 \mathrm{~mL})$, either malononitrile $\left(0.12 \mathrm{~g}, 1.86 \times 10^{-3} \mathrm{~mol}\right)$ or ethyl cyanoacetate $(0.21 \mathrm{~g}$, $1.86 \times 10^{-3} \mathrm{~mol}$ ) was added. The reaction mixture was heated under reflux for 3 hrs then poured onto a beaker containing an ice/water mixture and a few drops of hydrochloric acid. The solid product formed was collected by filtration and dried.

3-Cyano-4,6-diamino-2-oxo-1-imino-(4-methyl- $\omega$-phenylhydrazino-acetophenonylidieno)pyridine (12c) Green crystals (from ethanol). Yield: $0.51 \mathrm{~g}(70 \%)$, m.p. $80^{\circ} \mathrm{C}$; IR ( KBr ) v/cm ${ }^{-1}: 3480-3227\left(2 \mathrm{NH}_{2}\right.$, $2 \mathrm{NH}), 3028\left(\mathrm{CH}\right.$ aromatic), $2974\left(\mathrm{CH}_{3}\right), 2919\left(\mathrm{CH}_{2}\right), 2201(\mathrm{CN}), 1685(\mathrm{C}=\mathrm{O}), 1600(\mathrm{C}=\mathrm{C}) ;{ }^{1} \mathrm{H}-\mathrm{NMR}$ $\delta: 2.51\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 3.38\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 4.38,4.79,\left(2 \mathrm{~s}, 4 \mathrm{H}, 2 \mathrm{NH}_{2}\right), 6.48-8.19(\mathrm{~m}, 10 \mathrm{H}$, pyridine $\mathrm{H}-3$, $\left.\mathrm{C}_{6} \mathrm{H}_{5}, \mathrm{C}_{6} \mathrm{H}_{4}\right), 10.82,11.20(2 \mathrm{~s}, 2 \mathrm{H}, 2 \mathrm{NH}) ;{ }^{13} \mathrm{C}-\mathrm{NMR}: 23.7\left(\mathrm{CH}_{3}\right), 51.9\left(\mathrm{CH}_{2}\right), 80.7,89.3,114.9,125.3$ (pyridine C), $118.3(\mathrm{CN}), 126.4,128.0,129.3,129.6\left(\mathrm{C}_{6} \mathrm{H}_{5}\right), 170.9(\mathrm{C}=\mathrm{N}), 162.5(\mathrm{C}=\mathrm{O})$; Anal. Calcd. for $\mathrm{C}_{21} \mathrm{H}_{21} \mathrm{~N}_{7} \mathrm{O}$ (387.44): C, 65.10; H, 5.46; N, 25.30\%. Found: C, 64.21 ; H, 5.36; N, 25.38\%.

4-Amino-3-cyano-6-hydroxy-2-oxo-1-imino(4-methyl- $\omega$-phenyl-hydrazinoacetophenonylidieno) pyridine (12d). Yellowish green crystals (from ethanol). Yield: $0.58 \mathrm{~g}(80 \%) . \mathrm{Mp} 84-100{ }^{\circ} \mathrm{C}$; IR ( KBr ) $\mathrm{v} / \mathrm{cm}^{-1}$ : $3600-3185\left(\mathrm{OH}, \mathrm{NH}_{2}, 2 \mathrm{NH}\right), 3027\left(\mathrm{CH}\right.$ aromatic), $2975\left(\mathrm{CH}_{3}\right), 2919\left(\mathrm{CH}_{2}\right), 2206(\mathrm{CN}), 1684(\mathrm{C}=\mathrm{O})$, $1600(\mathrm{C}=\mathrm{C}) ;{ }^{1} \mathrm{H}-\mathrm{NMR} \delta: 2.50\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 3.39\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 4.34\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{NH}_{2}\right), 6.21(\mathrm{~s}, 1 \mathrm{H}$, pyridine $\mathrm{H}-3), 7.33-8.70\left(\mathrm{~m}, 9 \mathrm{H}, \mathrm{C}_{6} \mathrm{H}_{5}, \mathrm{C}_{6} \mathrm{H}_{4}\right), 10.84,11.22(2 \mathrm{~s}, 2 \mathrm{H}, 2 \mathrm{NH}), 12.64(\mathrm{~s}, 1 \mathrm{H}, \mathrm{OH}) ;{ }^{13} \mathrm{C}-\mathrm{NMR}: 23.6$ $\left(\mathrm{CH}_{3}\right), 51.6\left(\mathrm{CH}_{2}\right), 80.2,89.5,114.5,125.0$ (pyridine C), $118.1(\mathrm{CN}), 126.2,128.2,129.5,129.6$ $\left(\mathrm{C}_{6} \mathrm{H}_{5}\right), 164.8(\mathrm{C}=\mathrm{O}), 170.9(\mathrm{C}=\mathrm{N})$; Anal. Calcd. for $\mathrm{C}_{21} \mathrm{H}_{20} \mathrm{~N}_{6} \mathrm{O}_{2}$ (388.43): C, 64.93; H, 5.19; N, 21.63\%. Found: C, 65.42; H, 5.69; N, 21.13\%.

### 3.8. General Procedure for the Synthesis of 14a and 14b

To a solution of compound $\mathbf{6 a}\left(0.52 \mathrm{~g}, 2.12 \times 10^{-3} \mathrm{~mol}\right)$ in ethanol $(20 \mathrm{~mL})$ containing piperidine $(0.5 \mathrm{~mL})$, either acetylacetone $\left(0.21 \mathrm{~g}, 2.21 \times 10^{-3} \mathrm{~mol}\right)$ or ethyl acetoacetate $\left(0.27 \mathrm{~g}, 2.21 \times 10^{-3} \mathrm{~mol}\right)$ was added. The reaction mixture was heated under reflux for 3 hrs then poured onto a beaker containing an ice/water mixture and a few drops of hydrochloric acid. The solid product formed was collected by filtration and dried.

3-Cyano-4,6-diamethyl-2-oxo-1-imino-(4-methyl- $\omega$-hydrazinoaceto-phenonylidieno)pyridine
Brown crystals (from ethanol). Yield: $0.37 \mathrm{~g}(56 \%)$, m.p. $144^{\circ} \mathrm{C}$; IR ( KBr ) v/cm ${ }^{-1}: 3433-3229\left(\mathrm{NH}_{2}\right.$, NH ), 3027 ( CH aromatic), $2221(\mathrm{CN}), 1685(\mathrm{C}=\mathrm{O}), 1600(\mathrm{C}=\mathrm{C}) ;{ }^{1} \mathrm{H}-\mathrm{NMR}$ 8: 2.34, 2.54, $3.04(3 \mathrm{~s}, 9 \mathrm{H}$, $3 \mathrm{CH}_{3}$ ), $3.39\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 6.21(\mathrm{~s}, 1 \mathrm{H}$, pyridine $\mathrm{H}-3), 6.72-7.45\left(\mathrm{~m}, 4 \mathrm{H}, \mathrm{C}_{6} \mathrm{H}_{4}\right), 10.84,11.63(\mathrm{~s}, 2 \mathrm{H}$, 2NH); ${ }^{13} \mathrm{C}-\mathrm{NMR}: 16.0,19.2,24.3\left(3 \mathrm{CH}_{3}\right), 51.2\left(\mathrm{CH}_{2}\right), 80.0,88.3,115.3,123.9$ (pyridine C ), 116.8 $(\mathrm{CN}), 126.0,127.9,128.3,129.1\left(\mathrm{C}_{6} \mathrm{H}_{5}\right), 164.9(\mathrm{C}=\mathrm{O})$, $173.6(\mathrm{C}=\mathrm{N})$; Anal. Calcd. for $\mathrm{C}_{17} \mathrm{H}_{19} \mathrm{~N}_{5} \mathrm{O}$ (309.37): C, $66.00 ; \mathrm{H}, 6.19$; N, 22.63\%; found: C, $66.29 ; \mathrm{H}, 6.23 ; \mathrm{N}, 23.53 \%$.

3-Cyano-6-hydroxy-4-methyl-2-oxo-1-imino(4-methyl- $\omega$-hydrazino-acetophenonylidieno)pyridine
(14b). Brown crystals (from ethanol). Yield: $0.38 \mathrm{~g}(57 \%)$, m.p. $136^{\circ} \mathrm{C}$; IR ( KBr ) v/ $\mathrm{cm}^{-1}: 3549-3321$ $\left(\mathrm{OH}, \mathrm{NH}_{2}, \mathrm{NH}\right), 3027\left(\mathrm{CH}\right.$ aromatic), $2213(\mathrm{CN}), 1680(\mathrm{C}=\mathrm{O}), 1600(\mathrm{C}=\mathrm{C}) ;{ }^{1} \mathrm{H}-\mathrm{NMR}$ 8: 2.53, 3.13 $\left(3 \mathrm{~s}, 9 \mathrm{H}, 3 \mathrm{CH}_{3}\right), 3.39\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 6.22(\mathrm{~s}, 1 \mathrm{H}$, pyridine $\mathrm{H}-3), 6.73-7.39\left(\mathrm{~m}, 4 \mathrm{H}, \mathrm{C}_{6} \mathrm{H}_{4}\right), 10.85,11.33$ (s, 2H, 2NH), $12.21(\mathrm{~s}, 1 \mathrm{H}, \mathrm{OH}) ;{ }^{13} \mathrm{C}$-NMR: 16.1, 19.0, $24.3\left(3 \mathrm{CH}_{3}\right), 51.3\left(\mathrm{CH}_{2}\right), 80.2,88.3,115.3$, 152.1 (pyridine C), $116.6(\mathrm{CN}), 126.3,127.6,128.3,129.1\left(\mathrm{C}_{6} \mathrm{H}_{5}\right), 165.3(\mathrm{C}=\mathrm{O}), 173.8(\mathrm{C}=\mathrm{N})$; Anal. Calcd. for $\mathrm{C}_{16} \mathrm{H}_{17} \mathrm{~N}_{5} \mathrm{O}_{2}$ (311.34): C, 61.44; H, 5.50; N, 22.49\%; found: C, $61.23 ; \mathrm{H}, 5.67$; N, 22.30\%.

### 3.9. General Procedure for the Synthesis of 14c and 14d

To a solution of compound $\mathbf{6 b}\left(0.60 \mathrm{~g}, 1.86 \times 10^{-3} \mathrm{~mol}\right)$ in ethanol ( 20 mL ) containing piperidine $(0.5 \mathrm{~mL})$, either acetylacetone $\left(0.18 \mathrm{~g}, 1.86 \times 10^{-3} \mathrm{~mol}\right)$ or ethyl acetoacetate $\left(0.24 \mathrm{~g}, 1.86 \times 10^{-3} \mathrm{~mol}\right)$ was added. The reaction mixture was heated under reflux for 3 hrs then poured onto a beaker containing an ice/water mixture and a few drops of hydrochloric acid. The solid product formed was collected by filtration and dried.

3-Cyano-4,6-diamethyl-2-oxo-1-imino-(4-methyl- $\omega$-phenylhydrazino-acetophenonylidieno)pyridine
(14c) Yellowish brown crystals (from ethanol). Yield: $0.61 \mathrm{~g}(85 \%)$, m.p. $100^{\circ} \mathrm{C}$; IR ( KBr ) $\mathrm{v} / \mathrm{cm}^{-1}$ : 3450-3229 (2NH), 3027 (CH aromatic), $2213(\mathrm{CN}), 1680(\mathrm{C}=\mathrm{O}), 1600(\mathrm{C}=\mathrm{C}) ;{ }^{1} \mathrm{H}-\mathrm{NMR}$ 8: 2.34, 2.51, $3.01\left(3 \mathrm{~s}, 9 \mathrm{H}, 3 \mathrm{CH}_{3}\right), 3.37\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 6.21(\mathrm{~s}, 1 \mathrm{H}$, pyridine $\mathrm{H}-3)$, $6.50-8.41\left(\mathrm{~m}, 9 \mathrm{H}, \mathrm{C}_{6} \mathrm{H}_{5}, \mathrm{C}_{6} \mathrm{H}_{4}\right)$, $10.82,12.63(\mathrm{~s}, 2 \mathrm{H}, 2 \mathrm{NH}) ;{ }^{13} \mathrm{C}-\mathrm{NMR}: 16.3,19.1,24.3\left(3 \mathrm{CH}_{3}\right), 51.5\left(\mathrm{CH}_{2}\right), 80.3,88.6,115.0,127.7$, 155.8 (pyridine C), $116.5(\mathrm{CN}), 118.7,119.0,120.8,122.5,126.0,127.8,128.7,129.8\left(\mathrm{C}_{6} \mathrm{H}_{5}, \mathrm{C}_{6} \mathrm{H}_{4}\right)$, , $166.2(\mathrm{C}=\mathrm{O})$, $173.6(\mathrm{C}=\mathrm{N})$; Anal. Calcd. for $\mathrm{C}_{23} \mathrm{H}_{23} \mathrm{~N}_{5} \mathrm{O}$ (385.47): C, 71.66; H, 6.01; N, 18.16\%; found: C, 71.86; H, 5.98; N, 17.99\%.

## 3-Cyano-6-hydroxy-4-methyl-2-oxo-1-imino(4-methyl- $\omega$-phenyl-hydrazinoacetophenonylidieno)-

pyridine (14d). Yellowish brown crystals (from ethanol). Yield: 0.48 g ( $66 \%$ ). m.p. $96{ }^{\circ} \mathrm{C}$; IR (KBr) v/cm ${ }^{-1}: 3600-3220(\mathrm{OH}, 2 \mathrm{NH}), 3027(\mathrm{CH}$ aromatic), $2211(\mathrm{CN}), 1682(\mathrm{C}=\mathrm{O}), 1600(\mathrm{C}=\mathrm{C})$; ${ }^{1} \mathrm{H}-\mathrm{NMR} \delta: 2.51,3.00\left(2 \mathrm{~s}, 6 \mathrm{H}, 2 \mathrm{CH}_{3}\right), 3.40\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 6.54(\mathrm{~s}, 1 \mathrm{H}$, pyridine $\mathrm{H}-3), 7.34-8.40(\mathrm{~m}$, $\left.9 \mathrm{H}, \mathrm{C}_{6} \mathrm{H}_{5}, \mathrm{C}_{6} \mathrm{H}_{4}\right), 9.62,10.82(2 \mathrm{~s}, 2 \mathrm{H}, 2 \mathrm{NH}), 12.65(\mathrm{~s}, 1 \mathrm{H}, \mathrm{OH}) ;{ }^{13} \mathrm{C}-\mathrm{NMR}: 16.3,19.4,24.6\left(3 \mathrm{CH}_{3}\right)$, $51.5\left(\mathrm{CH}_{2}\right), 80.8,88.4,116.5,155.4$ (pyridine C), $116.5(\mathrm{CN}), 118.8,119.0,120.3,121.7,126.0$, 127.8, 128.9, $129.0\left(\mathrm{C}_{6} \mathrm{H}_{5}, \mathrm{C}_{6} \mathrm{H}_{4}\right)$, , $166.4(\mathrm{C}=\mathrm{O})$, $173.9(\mathrm{C}=\mathrm{N})$; Anal. Calcd. for $\mathrm{C}_{22} \mathrm{H}_{21} \mathrm{~N}_{5} \mathrm{O}_{2}$ (387.44): C, 68.20; H, 5.46; N, 18.07\%; Found: C, 68.49; H, 5.84; N, 17.79\%.

2-Acetonitrilo-5-(4-methylphenyl)-1,3,4-oxadiazine (15). A solution of compound 3 ( 1.00 g , $\left.3.39 \times 10^{-3} \mathrm{~mol}\right)$ in sodium ethoxide ( 50 mL ) was heated under reflux for 3 hrs then poured onto a beaker containing an ice/water mixture and a few drops of hydrochloric acid. The formed solid product was collected by filtration and dried to give pale brown crystals (from ethanol). Yield $0.39 \mathrm{~g}(54 \%)$, m.p. $190-193{ }^{\circ} \mathrm{C}$; IR (KBr) v/cm ${ }^{-1}: 3028\left(\mathrm{CH}\right.$ aromatic), $2998\left(\mathrm{CH}_{3}\right), 2922,2854\left(2 \mathrm{CH}_{2}\right), 2210(\mathrm{CN})$, $1673(\mathrm{C}=\mathrm{N}), 1609(\mathrm{C}=\mathrm{C}) ;{ }^{1} \mathrm{H}-\mathrm{NMR} \delta: 2.51\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 3.35,4.42\left(2 \mathrm{~s}, 4 \mathrm{H}, 2 \mathrm{CH}_{2}\right), 7.00-8.40(\mathrm{~m}, 4 \mathrm{H}$, $\left.\mathrm{C}_{6} \mathrm{H}_{4}\right)$; ${ }^{13} \mathrm{C}$-NMR: $24.4\left(\mathrm{CH}_{3}\right), 20.3\left(\mathrm{CH}_{2}\right), 64.3$ (oxadiazine $\mathrm{CH}_{2}$ ), $116.9(\mathrm{CN}), 126.0,127.8,128.9$, 129.0, $133.1\left(\mathrm{C}_{6} \mathrm{H}_{5}\right)$, $170.5(\mathrm{C}=\mathrm{N})$; Anal. Calcd. for $\mathrm{C}_{12} \mathrm{H}_{11} \mathrm{~N}_{3} \mathrm{O}$ (213.24): C, 67.59; H, 5.19; N , $19.70 \%$; found: C, 67.70 ; H, 5.38 ; N, 19.88\%.

2-( $\alpha$-Benzalacetonitrilo)-5-(4-methylphenyl)-1,3,4-oxadiazine hydrochloride (16). To a solution of compound $3\left(2.00 \mathrm{~g}, 6.79 \times 10^{-3} \mathrm{~mol}\right)$ in ethanol ( 30 mL ) containing piperidine $(0.5 \mathrm{~mL})$, benzaldehyde $\left(0.72 \mathrm{~g}, 6.79 \times 10^{-3} \mathrm{~mol}\right)$ was added. The reaction mixture was heated under reflux for 3 hrs then poured onto a beaker containing an ice/water mixture with a few drops of hydrochloric acid. The formed solid product was collected by filtration and dried to afford brown crystals (from ethanol). Yield: $1.35 \mathrm{~g}(58 \%)$, m.p. $100{ }^{\circ} \mathrm{C}$; IR $(\mathrm{KBr}) \mathrm{v} / \mathrm{cm}^{-1}: 3050\left(\mathrm{CH}\right.$ aromatic), $3029\left(\mathrm{CH}_{3}\right), 2921\left(\mathrm{CH}_{2}\right)$,
$2830(\mathrm{CH}), 2216(\mathrm{CN}), 1604(\mathrm{C}=\mathrm{C})$ ) ${ }^{1} \mathrm{H}-\mathrm{NMR} \delta: 2.51\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 4.22\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 5.16(\mathrm{~s}, 1 \mathrm{H}$, $=\mathrm{CH}), 7.35-8.02\left(\mathrm{~m}, 9 \mathrm{H}, \mathrm{C}_{6} \mathrm{H}_{5}, \mathrm{C}_{6} \mathrm{H}_{4}\right) ;{ }^{13} \mathrm{C}-\mathrm{NMR}: 24.6\left(\mathrm{CH}_{3}\right), 63.8$ (oxadiazine $\left.\mathrm{CH}_{2}\right), 116.7(\mathrm{CN})$, $118.2(\mathrm{CH}=\mathrm{C}), 120.4,121.9,124.5,126.2,127.4,127.9,129.2,133.9\left(\mathrm{C}_{6} \mathrm{H}_{4}, \mathrm{C}_{6} \mathrm{H}_{5}\right), 144.2(\mathrm{CH}=\mathrm{C})$, 171.5, $173.9(2 \mathrm{C}=\mathrm{N})$; Anal. Calcd. for $\mathrm{C}_{19} \mathrm{H}_{15} \mathrm{~N}_{3} \mathrm{O}$ (301.34): C, 75.73; H, 5.02 ; N, 13.94\%; found: C, 75.52; H, 4.91; N, 14.31\%.

2-( $\alpha$-Phenylhydrazoacetonitrilo)-5-(4-methylphenyl) 1,3,4-oxadiazine (17). To a cold solution $\left(0-5{ }^{\circ} \mathrm{C}\right.$ ) of compound $15\left(1.00 \mathrm{~g}, 3.39 \times 10^{-3} \mathrm{~mol}\right)$, in ethanol ( 50 mL ) was added with continuous stirring benzenediazonium chloride ( $3.39 \times 10^{-3} \mathrm{~mol}$ ) [which was prepared by dissolving sodium nitrite $\left(0.35 \mathrm{~g}, 5.09 \times 10^{-3} \mathrm{~mol}\right)$ in water, 2 mL was added to a cold solution of aniline $\left(0.31 \mathrm{~g}, 3.39 \times 10^{-3} \mathrm{~mol}\right)$ containing the appropriate amount of hydrochloric acid and with continuous stirring]. The solid product formed after adjusting the pH 6 and stirring was collected by filtration and dried to give reddish brown crystals (from ethanol and dimethylformamide). Yield: $0.96 \mathrm{~g}(89 \%)$, m.p. $130{ }^{\circ} \mathrm{C}$; IR $(\mathrm{KBr}) \mathrm{v} / \mathrm{cm}^{-1}: 3300-3215(\mathrm{NH}), 3057\left(\mathrm{CH}\right.$ aromatic), $3030\left(\mathrm{CH}_{3}\right), 2920\left(\mathrm{CH}_{2}\right), 2219(\mathrm{CN}), 1679$ $(\mathrm{C}=\mathrm{N}), 1603(\mathrm{C}=\mathrm{C}) ;{ }^{1} \mathrm{H}-\mathrm{NMR} \delta: 2.52\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 3.38\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 7.21-8.82\left(\mathrm{~m}, 9 \mathrm{H}, \mathrm{C}_{6} \mathrm{H}_{5}, \mathrm{C}_{6} \mathrm{H}_{4}\right)$, $10.22(\mathrm{~s}, 1 \mathrm{H}, \mathrm{NH})$. ). ${ }^{13} \mathrm{C}$-NMR: $24.3\left(\mathrm{CH}_{3}\right), 63.0\left(\right.$ oxadiazine $\left.\mathrm{CH}_{2}\right), 116.5(\mathrm{CN}), 121.5,121.9,124.0$, 123.5, 126.7, 127.0, 127.6, 129.2, $133.0\left(\mathrm{C}_{6} \mathrm{H}_{4}, \mathrm{C}_{6} \mathrm{H}_{5}\right), 166.9,170.5,173.9(3 \mathrm{C}=\mathrm{N})$. Anal. Calcd. for $\mathrm{C}_{18} \mathrm{H}_{15} \mathrm{~N}_{5} \mathrm{O}$ (317.35): C, 68.12; H, 4.76; N, 22.06\%; found: C, 67.92; H, 4.91; N, 21.25\%.

## 4. Conclusions

In this work cyanoacetyl hydrazine (1) reacted with $\alpha$-haloketone 2 to afford the $\alpha$-bromohydrazide-hydrazone derivative 3 . The latter compound was used in a series of heterocyclic transformations to give compounds that were tested as antitumor agents. The 2-hydroxypyridine derivatives $\mathbf{1 4 b}$ showed the highest inhibitory activity.

## Acknowledgments

R. M. Mohareb expressed his deepest thank to the Alexander von Humboldt Foundation in Bonn for its financial support through a fellowship in Erlangen University during Summer 2009 and for the completion of this work.

## References and Notes

1. Bharti, S.K.; Nath, G.; Tilak, R.; Singh, S.K. Synthesis, anti-bacterial and anti-fungal activities of some novel Schiff bases containing 2,4-disubstituted thiazole ring. Eur. J. Med. Chem. 2010, 45, 651-660.
2. Loncle, C.; Brunel, J.M.; Vidal, N.; Dherbomez, M.; Letourneux, Y. Synthesis and antifungal activity of cholesterol-hydrazone derivatives. Eur. J. Med. Chem. 2004, 39, 1067-1071.
3. Garoufalias, S.P.; Pouli, N.; Marakos, P.; Ladas, A.C. Synthesis antimicrobial and antifungal activity of some new 3 -substituted derivatives of 4-(2,4-dichlorophenyl)-5-adamantyl-1 H -1,2,4triazole. Farmaco 2002, 57, 973-977.
4. Vicini, P.; Zani, F.; Cozzini P.; Doytchinova, I. Hydrazones of 1,2-benzisothiazole hydrazides: synthesis, antimicrobial activity and QSAR investigations. Eur. J. Med. Chem. 2002, 37, 553-567.
5. Popp, F.D. Potential anticonvulsant. XII. anticonvulsant activity of some aldehyde derivatives. Eur. J. Med. Chem. 1989, 24, 313-316.
6. Sridhar, S.K.; Pandeya, S.N.; Stables J.P.; Atmakuru, R. Anticonvulsant activity of hydrazones, Schiff and Mannich bases of isatin derivatives Eur. J. Pharm. Sci. 2002, 16, 129-132.
7. Küçükgüzel, S.G.; Mazi, A.; Sahin, F.; Öztürk, S.; Stables, J.P. Synthesis and biological activities of diflunisal hydrazide-hydrazones. Eur. J. Med. Chem. 2003, 38, 1005-1013.
8. Todeschini, A.R.; Miranda, A.L.P.; Silva, K.C.M.; Parrini, S.C.; Barreiro, E. Synthesis and evaluation of analgesic, antiinflammatory and antiplatelet properties of new 2-pyridylarylhydrazone derivatives. Eur. J. Med. Chem. 1998, 33, 189-199.
9. Gaston, M.A.; Dias, L.R.S.; Freitas, A.C.C.; Miranda, A.L.P.; Barreiro, E. Synthesis and Analgesic Properties of New 4-Arylhydrazone 1H-Pyrazolo[3,4-b] Pyridine Derivatives. J. Pharm. Acta Helv. 1996, 71, 213-219.
10. Melnyk, P.; Leroux, V.; Sergheraert, C.; Grellier, P. Design, synthesis and in vitro antimalarial of an acylhydrazone library. Bioorg. Med. Chem. Lett. 2006, 16, 31-35.
11. Küçükgüzel, S.G.; Rollas, S., Küçükgüzel, I.; Kiraz, M. Synthesis and antimycobacterial activity of some coupling products from 4-aminobenzoic acid hydrazones. Eur. J. Med. Chem. 1999, 34, 1093-1100.
12. Kaymakçıoğlu, B.K.; Rollas, S. Synthesis, characterization and evaluation of antituberculosis activity of some hydrazones. Farmaco 2002, 57, 595-599.
13. Terzioglu, N.; Gursoy, A. Synthesis and anticancer evaluation of some new hydrazone derivatives of 2,6-dimethylimidazo[2,1-b][1,3,4]thiadiazole-5-carbohydrazide. Eur. J. Med. Chem. 2003, 38, 781-786.
14. Sondhi, S.M.; Dinodia, M., Kumar, A. Synthesis, anti-inflammatory and analgesic activity evaluation of some amidine and hydrazone derivatives Bioorg. Med. Chem. 2006, 14, 4657-4663.
15. Boga, C.; Fiume, L.; Baglioni, M.; Bertucci, C.; Farina, C.; Kratz, F.; Manerba, M.; Naldi, M.; Stefano, G. Characterisation of the conjugate of the (6-maleimidocaproyl)hydrazone derivative of doxorubicin with lactosaminated human albumin by ${ }^{13} \mathrm{C}$ NMR spectroscopy. Eur. J. Pharm. Sci. 2009, 38, 262-269.
16. Garnett, M.C. Targeted drug conjugates: principles and progress. Adv. Drug Deliv. Rev. 2001, 53, 171-216.
17. Rodrigues, P.C.; Scheuermann, K.; Stockmar, C.; Maier, G.; Fiebig, H.; Unger, C.; Mülhaupt, R.; Kratz, F. Synthesis and In vitro efficacy of acid-Sensitive poly(ethylene glycol) paclitaxel conjugates. Bioorg. Med. Chem. Lett. 2003, 13, 355-360.
18. El-Sabbagh, O.I.; Rady, H.M. Synthesis of new acridines and hydrazones derived from cyclic $\beta$-diketone for cytotoxic and antiviral evaluation. Eur. J. Med. Chem. 2009, 44, 3680-3686.
19. Krakovicova, H.; Etrych, T.; Ulbrich, K. HPMA-based polymer conjugates with drug combination. Eur. J. Pharm. Sci. 2009, 37, 405-412.
20. Zhang, H.Z.; Drewe, J.; Tseng, B.; Kasibhatla, S.; Cai, S.X. Discovery and SAR of indole-2carboxylic acid benzylidene-hydrazides as a new series of potent apoptosis inducers using a cellbased HTS assay. Bioorg. Med. Chem. 2004, 12, 3649-3655.
21. Vicini, P.; Incerti, M.I.; Colla, P.L.; Loddo, R. Anti-HIV evaluation of benzo[d]isothiazole hydrazones. Eur. J. Med. Chem. 2009, 44, 1801-1807.
22. Gürsoy, E.; Güzeldemirci, N.U. Synthesis and primary cytotoxicity evaluation of new imidazo[2,1-b]thiazole derivatives. Eur. J. Med. Chem. 2007, 42, 320-326.
23. Hao, J.J.; Xu, Y.; Geng,C.; Liu, W.Y.; Wang, E.; Gong, Z.; Ulbrich, N. Purification of $\alpha$-Sarcin and an Antifungal Protein from Aspergillus giganteusby Blue Sepharose CL-6B Affinity Chromatography. Protein Expr. Purif. 1998, 14, 295-301.
24. Ke, S.Y.; Qian, X.H.; Liu, F.Y.; Wang, N.; Yang, Q.; Zhong, L. Novel 4H-1,3,4-oxadiazin-5(6H)ones with hydrophobic and long alkyl chains: Design, synthesis, and bioactive diversity on inhibition of monoamine oxidase, chitin biosynthesis and tumor cell. Eur. J. Med. Chem. 2009, 44, 2113-2121.

Sample Availability: Samples of the compounds 4a-17 are available from the authors.
© 2010 by the authors; licensee MDPI, Basel, Switzerland. This article is an Open Access article distributed under the terms and conditions of the Creative Commons Attribution license (http://creativecommons.org/licenses/by/3.0/).

