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Synthesis, characterization and molecular docking studies of thiouracil derivatives as potent thymidylate synthase inhibitors and potential anticancer agents

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Abstract Thymidylate synthase (TS), one of folatedependent enzymes, is a key and well-recognized target for anticancer agents. In this study, a series of 6-aryl-5cyano thiouracil derivatives were designed and synthesized in accordance with essential pharmacophoric features of known TS inhibitors. Nineteen compounds were screened in vitro for their anti-proliferative activities toward HePG-2, MCF-7, HCT-116, and PC-3 cell lines. Compounds 21c, 21d, and 24 exhibited high anti-proliferative activity, comparable to that of 5-fluorouracil. Additionally, ten compounds with potent anti-proliferative activities were further evaluated for their ability to inhibit TS enzyme. Six compounds (21_b, 21_c, 21_d, 22, 23 and 24) demonstrated potent dose-related TS inhibition with IC₅₀ values ranging from 1.57 to $3.89 \,\mu$ M. The in vitro TS activity results were consistent with those of the cytotoxicity assay where the most potent anti-proliferative compounds of the series showed good TS inhibitory activity

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comparable to that of 5-fluorouracil. Furthermore, molecular docking studies were carried out to investigate the binding pattern of the designed compounds with the prospective target, TS (PDB-code: 1JU6).

Keywords Anticancer · Thymidylate synthase · 6-Aryl-5cyano thiouracil · Docking

Introduction

Cancer is a prime public health problem all over the world [1]. It initiates when cells in a part of the body start to grow without control [2]. Cancer is the second main cause of death in the USA [3]. Today, millions of people are living with cancer [3]. Approximately 13% of all deaths worldwide, numbering 8.2 million people annually, are due to cancer. Currently, more than 100 cancer types exist and each needs unique diagnosis and treatment [4,5].

Thymidylate synthase (TS), a folate-dependent enzyme, is a key well-recognized target for anticancer agents. Due to its crucial role in DNA synthesis, it is an important enzyme in all body functions [6]. The main role of TS lies in its dual function; it acts as a catalytic enzyme and as an effective regulator of its own expression by binding and inactivating its own RNA. With regard to its catalytic function, it catalyzes the reductive methylation of 2'-deoxyuridine-5'-monophosphate (dUMP) to 2'-deoxythymidine-5'-monophosphate (dTMP) [7]. The regulatory role of TS may be implicated in the synthesis of key proteins that regulate the apoptotic process [8].

TS inhibitors bind to TS through two main essential moieties: pyrimidine and glutamate. The pyrimidine moiety binds to a pyrimidine-binding site, while the glutamate moiety binds to a folate-binding site [9,10].



Fig. 1 Structures and basic pharmacophoric features of some reported TS inhibitors

5-Fluorouracil (5-FU) 4 and its modified analogue, 5fluorothiouracil 5, are widely used anticancer agents [11-13]. These drugs can be inserted into RNA and/or DNA resulting in interference with the maturation of nuclear RNA and/or DNA. However, the main mechanism of action of fluorouracil is its conversion into 5-fluoro-2'-deoxy-5'-monophosphate (FdUMP) resulting in inhibition of TS and DNA synthesis [10]. 5-FU treatment has various adverse effects, including (i) toxic side effects due to its misincorporation into RNA, (ii) resistance often developed due to a decrease in kinases cellular levels required for metabolic activation, and (iii) as a single antineoplastic agent, it has a narrow range of activity [9]. Macchia et al. [14] designed and synthesized compounds 6 and 7, modified analogues of 5-fluorouracil and 5-fluorothiouracil, respectively. These compounds were screened for anti-proliferating in vitro and in vivo anti-tumor activity. These compounds showed promising anticancer activity with IC₅₀ values of 7.9 and $1.7 \,\mu$ M, respectively.

The crystal structures of TS from mammalian and bacterial sources provided key information [15] to design and synthesize new TS inhibitors, e.g., raltitrexed 1 [16], pemetrexed 2 [17] and nolatrexed 3 [18]. Raltitrexed 1 and pemetrexed 2 contain glutamate moieties that can bind to the enzyme at both folate- and pyrimidine-binding sites. The new non-

classical generation of TS-directed inhibitors such as nolatrexed **3** is lacking a negatively charged glutamate side chain. The design of this TS inhibitor is based on computer-assisted design using high-resolution X-ray crystallography [19–22].

Based on the previously mentioned findings, we decided to design and synthesize a novel series of thiouracils incorporating a nitrile group instead of the fluoro group. The synthesized compounds were designed to contain a pyrimidine moiety, the main and essential pharmacophoric feature of TS inhibitors, with the major target of developing agents with potential anticancer activity that could be more effective than 5-fluorouracil.

Rationale of molecular design

Many reported facts were clarified before the chemical synthesis of our target compounds. The first fact was the presence of a pyrimidine nucleus is essential as a basic pharmacophoric feature for TS inhibition [23]. The second fact was modifications at the fifth and sixth positions of 5-fluorouracil and 5-fluorothiouracil led to the generation of many TS inhibitors such as compounds **6** and **7** with potent anticancer activities [14] (Fig. 1). The third fact was the existence



Fig. 2 Rationale of molecular design of new anticancer agents

of new non-classical generation of TS inhibitors without a negatively charged glutamate side chain [24]. Based on all this information, ligand-based drug design approach was achieved to improve chemotherapeutic activity of these inhibitors.

The main core of our molecular design rationale comprises chemical modifications on 5-fluorothiouracil at four different positions (Fig. 2). The first position is C5, with the replacement of a fluoro group with isosteric nitrile group. This modification was based on several bioisosteric considerations: (i) The nitrile group has a crucial role as hydrogen bond acceptor [25,26]. Many publications reported that nitrile protrudes into strict active sites to form an important polar hydrogen bonding interaction [27]; (ii) the nitrile may increase the polar character of the aromatic π -system [28]; (iii) moreover, the nitrile group may not engage in specific interactions, but may enhance pharmacokinetic properties. Solubility studies of farnesyltransferase inhibitors revealed that nitrile-containing members were nearly tenfold more soluble than the corresponding bromo analogues [29].



Scheme 1 General procedure for synthesis of target compounds 11a,b

The second position is C6 through phenyl substitution. The insertion of a phenyl group may increase binding affinity since the synthase enzyme is very large and contains a deep pocket for the binding of both substrate and cofactor [30]. The third and fourth positions are N-3 heteroatom and the SH group with three pathways of modifications: (i) chemical substitutions on the SH moiety with different alkyl and aralkyl moieties, (ii) chemical substitutions on both N-3 atom and SH group with different alkyl and aralkyl moieties, and (iii) using N-3 and SH groups to form five- or six-membered rings.

This rationale of molecular design allowed the generation of new hybrid structures with different biological isosteres. The wide variety of chemical modifications may help us to study the SAR of these novel compounds as anticancer agents. The binding mode of 5-fluorouracil with the active site of TS has been unequivocally elucidated [31,32]. Based on this information, we carried out molecular docking studies to confirm the binding manner as well as the inhibitory activity of the synthesized compounds toward TS.

Results and discussion

Chemistry

The sequence of reactions used in the synthesis of the title compounds is shown in Schemes 1–4. The starting key compounds $11_{a,b}$ were synthesized via a one-pot cyclocondensation reaction where equivalent molar quantities of aromatic aldehydes, ethyl cyanoacetate and thiourea reacted in absolute ethanol in the presence of anhydrous K₂CO₃ followed by neutralization with acetic acid [33] (Scheme 1).

Compound 11_a reacted with commercially available 1,2dibromoethane in the presence of anhydrous K₂CO₃ and tetrabutylammonium bromide (TBAB) in dry THF to afford cyclo-substitution product **12**. Compound **11**_b underwent a nucleophilic displacement reaction with acetyl chloride to give compound 13. Furthermore, reaction of 11_a with methylene chloride produced 2,2'-[methylenebis(sulfanediyl)]bis [4-(4-methoxyphenyl)-6-oxo-1,6-dihydropyrimidine-5-carb onitrile] 14 (Scheme 2).

S-Alkylation reactions were carried out between equimolar amounts of key compounds $11_{a,b}$ and appropriate commercially available halo derivatives, namely allyl bromide, chloroacetic acid, ethyl chloroacetate, diethyl bromomalonate and ethyl chloroformate in the presence of anhydrous K₂CO₃ and tetrabutylammonium bromide in dry THF to give the corresponding thiouracil derivatives **15–19**, respectively. While the reaction of $11_{a,b}$ with two molar equivalent of benzyl chloride gave di-benzylated carbonitriles **20**_{a,b} (Scheme 3).

Since Mannich reaction is a convenient method for the introduction of an amino methylene fragment in compounds bearing an acidic proton, we found compounds $11_{a,b}$ are suitable substrates for such amino methylation process. Reaction of compounds $11_{a,b}$ with one equivalent of an appropriate primary amine, namely 4-methoxyaniline, p-toluidine, 2-amino-pyrimidine, iso-butylamine and 2-naphthylamine, with excess of formaldehyde (37%) in dioxane produced the corresponding target compounds 21-24, respectively (Scheme 4). The structures of all newly synthesized novel compounds were confirmed by spectral and elemental analyses.

Biological evaluation

In vitro anti-proliferative activity

In vitro evaluation of the newly synthesized compounds for their anti-proliferative activities was carried out against four human cancer cell lines, namely hepatocellular carcinoma (HePG-2), breast carcinoma (MCF-7), colorectal carcinoma (HCT-116) and prostate cancer (PC-3). The biological eval-



Scheme 2 General procedure for synthesis of target compounds 12–14

uation was performed using a sulforhodamine B (SRB) colorimetric assay as described by Skehan et al. [34]. The cytotoxicity results were expressed as median inhibitory concentration (IC_{50}) values for the synthesized compounds using drug 5-fluorouracil as reference (Table 1).

The screened compounds showed different levels of cytotoxicity ranging from excellent, to moderate, to weak activities against the above 4 cell lines. In general, the best performers are compounds 21_c , 21_d and 24, of which compound 24 was the most potent one being 1.33, 1.28, 1.17, and 1.98 times more active than 5-fluorouracil against HePG-2 (IC₅₀ = $6.4 \pm 0.14 \,\mu$ M), MCF-7 (IC₅₀ = $5.9 \pm 0.11 \,\mu$ M), HCT-116 (IC₅₀ = $7.9 \pm 0.14 \,\mu$ M) and PC-3 (IC₅₀ = $5.2 \pm 0.09 \,\mu$ M), respectively.

Compounds 13, 17_a, 17_b, 21_a, 21_b, 22 and 23 were found to possess moderate anti-proliferative activities with IC₅₀ values ranging from 8.7 to $28.7 \,\mu$ M. The rest of the compounds were weaker inhibitors.

In vitro assay of thymidylate synthase (TS) activity

Compounds exhibiting anti-proliferative activities $(13, 17_a, 17_b, 21_a, 21_b, 21_c, 21_d, 22, 23$ and 24) were further evaluated to determine their inhibitory activities against thymidylate synthase (TS). The kinase activity of TS was measured spectrophotometrically according to reported procedure described by Wahba and Friedkin [35]. 5-Fluorouracil was used as a positive control in this assay. The results were reported as enzymatic inhibition percentage (%) and a 50%

inhibition concentration values (IC₅₀) calculated from the concentration—inhibition response curve and are summarized in Table 1 and Fig. 3.

The tested compounds exhibited good to moderate inhibitory activity with IC₅₀ values ranging from 1.57 to 14.04 μ M, which were consistent with that of their in vitro cytotoxicity activities. 5-Fluorouracil as a reference compound showed 73.40% TS inhibition. Some of the test compounds exhibited higher activities than the reference standard (5-fluorouracil). Compounds **21**_d and **23** demonstrated potent inhibition, more than 80% for the TS activity with IC₅₀ values of 1.57 ± 0.08 and 2.5 ± 0.65 μ M, respectively. In addition, significant inhibition above 70% was exhibited by several investigated compounds, namely **17**_b, **21**_b, **22** and **24**, with IC₅₀ values of 5.13 ± 0.11, 3.66 ± 0.44, 3.51 ± 0.24 and 3.89 ± 0.42 μ M, respectively.

Molecular docking

In our work, docking of the synthesized compounds and reference ligands, 5-FU and LYA (co-crystallized ligand) with the TS crystal structure (ID: 1JU6), was carried to predict the binding mode, affinity and preferred orientation of each docking pose. The results of docking study are reported as TS binding free energy (ΔG) (Table 2).

The key binding site of TS has been reported [31,36] and involves amino acid residues of Phe225, Leu221 and Ile108. As reported, 5-FU was found to interact with two important



Scheme 3 General procedure for synthesis of target compounds ${\bf 15}_{a,b}{-}{\bf 20}_{a,b}$

amino acid residues of the active site (Leu221 and Phe225) [31].

The proposed binding mode of 5-FU showed an affinity value of -60.20 kcal/mol (RMSD = 1.20). The NH group at the 3-position was involved in a hydrogen bonding interaction with the carbonyl backbone of Ser216, the C=O group at the 2-position formed a hydrogen bond with side chain of

Ser216, and the C=O group at the 4-position was involved in a hydrogen bonding interaction with Asn226. The aromatic moiety was involved in aromatic stacking interactions with Phe225 and Leu221 (Fig. 4).

The proposed binding mode of the re-docked cocrystallized ligand (LYA) showed affinity value of -90.32kcal/mol. The side chain NH₂ group in position two was



Scheme 4 General procedure for synthesis of target compounds $21_{a-d}-24$

stabilized by a hydrogen bonding interaction with carbonyl backbone of Ala312, besides forming a water-mediated interaction with Asn112. Furthermore, the NH group at the 3-position formed a hydrogen bond with Asp218. The phenyl ring was stabilized by hydrophobic interactions with Met311, Leu221, Ile108 and aromatic stacking interaction with Trp109 and Phe225 (Fig. 5).

Overall, it was found that most of the studied compounds are expected to have the same binding mode of 5-FU and LYA. The designed molecules showed binding energies ranging from -37.24 to -83.71 kcal/mol (Table 2).

The proposed binding mode of compound **24** showed an affinity value of -83.71 kcal/mol engaging in several interactions with the protein. The bicyclic moiety was involved in aromatic stacking interactions with Phe80 and Phe225. Moreover, the bicyclic moiety was located in the hydropho-

bic pocket formed by Ile108 and Leu221. The phenyl ring was bonded by cationic– π interactions with Arg50 and Tyr258. The carbonyl group at the 6-position of pyrimidine moiety was anchored by hydrogen bonding interactions with Gln214 and Asn226. The cyano group at the 5-position of pyrimidine moiety is expected to participate in a hydrogen bonding interaction with Ser216. Due to the presence of extra hydrogen bonding and favorable interactions, compound **24** is predicted to have higher affinity toward the receptor compared over other compounds (Fig. 6).

On the other hand, compound 17_b exhibited promising TS inhibitory activity (IC₅₀ = $5.13 \pm 0.11 \,\mu\text{M}$ and TS inhibition% = 79.28). Docking energy data of the compound 17_b indicated that the ethyl acetate moiety is important for increasing the affinity toward the active site. Moreover, it was found that the binding mode of compound 17_b is predicted

Table 1In vitroanti-proliferative activitiestoward HePG-2, MCF-7,HCT-116 and PC-3 cell linesand thymidylate synthase (TS)

Comp.	$IC_{50}(\mu M)^a$		TS			
	HePG-2	MCF-7	HCT-116	PC-3	$\overline{IC_{50}(\mu M)^a}$	Inhibition %
12	35.2 ± 0.72	28.4 ± 0.59	43.7 ± 0.69	42.2 ± 0.92	NT ^c	NT ^c
13	27.5 ± 0.33	16.5 ± 0.31	15.8 ± 0.26	13.1 ± 0.27	14.04 ± 0.63	49.37
14	35.4 ± 0.45	NA ^b	46.2 ± 0.83	NA ^b	NT ^c	NT ^c
15 _a	30.9 ± 0.52	46.1 ± 1.06	NA ^b	38.9 ± 0.73	NT ^c	NT ^c
15 _b	NA ^b	NA ^b	14.8 ± 0.32	NA ^b	NT ^c	NT ^c
16	38.8 ± 0.65	48.2 ± 0.77	36.7 ± 0.73	57.9 ± 1.27	NT ^c	NT ^c
17 _a	25.5 ± 0.04	16.3 ± 0.34	19.2 ± 0.34	18.9 ± 0.35	5.49 ± 0.29	68.10
17 _b	16.8 ± 0.03	8.7 ± 0.16	16.1 ± 0.30	10.6 ± 0.19	5.13 ± 0.11	79.28
18	NA ^b	NA ^b	NA ^b	NA ^b	NT ^c	NT ^c
19	29.2 ± 0.61	29.0 ± 049	28.2 ± 0.53	35.6 ± 0.60	NT ^c	NT ^c
20 _a	NA ^b	NA ^b	45.1 ± 0.19	NA ^b	NT ^c	NT ^c
20 _b	NA ^b	48.3 ± 0.33	NA ^b	NA ^b	NT ^c	NT ^c
21 _a	28.6 ± 0.84	15.4 ± 0.33	28.5 ± 0.48	19.5 ± 0.73	5.3 ± 0.41	57.09
21 _b	24.2 ± 0.53	9.7 ± 0.17	10.9 ± 0.20	27.3 ± 0.49	3.66 ± 0.44	70.41
21 _c	11.1 ± 0.21	8.1 ± 0.13	10.00 ± 0.23	7.4 ± 0.15	2.14 ± 0.22	66.71
21 _d	10.8 ± 0.22	13.8 ± 0.20	9.8 ± 0.15	11.9 ± 0.39	1.57 ± 0.08	83.09
22	16.8 ± 0.34	14.4 ± 0.34	17.2 ± 0.34	13.8 ± 0.27	3.51 ± 0.24	75.97
23	13.9 ± 0.25	16.6 ± 0.38	19.8 ± 0.35	15.0 ± 0.25	2.5 ± 0.65	81.43
24	6.4 ± 0.14	5.9 ± 0.11	7.9 ± 0.14	5.2 ± 0.09	3.89 ± 0.42	76.93
5-FU	8.5 ± 0.13	7.6 ± 0.15	9.3 ± 0.19	10.3 ± 0.18	1.84 ± 0.014	73.40

 a IC_{50} values are the mean \pm S.D. of three separate experiments

^b NA: compounds with IC₅₀ values > 50 μ M

^c NT not tested





to be the same as that of compound **24**. In addition, the ethyl group of the compound 17_b is stabilized by hydrophobic interactions with Ile108, Leu221 and Phe225. However, the substitution with carbonyl group can decrease the affinity for the receptor, where the carbonyl groups are surrounded by unfavorable interactions with nonpolar groups (Ile108 and Met311) (Fig. 7).

Compound 21_d showed binding mode with TS similar to the general pattern observed by compound 24. The hydrogen bonding interactions and aromatic stacking interactions were maintained. Compound 24 and 21_d can fit nicely inside the binding pocket (Fig. 8). **Fig. 4** Predicted binding mode of 5-FU at the TS binding site. H-bonds (*yellow dotted lines*), hydrogen atom (*gray*), nitrogen atom (*blue*), and oxygen atom (*red*). (Color figure online)

Fig. 5 Predicted binding mode of co-crystallized ligand (LYA) at the TS binding site. H-bonds (*yellow dotted lines*), hydrogen atom (*gray*), nitrogen atom (*blue*), and oxygen atom (*red*). (Color figure online)



Structure-activity relationships (SAR)

As outlined earlier, we first explored the effect of the chemical substitutions on the SH group with different alkyl and aralkyl moieties. The decreased IC_{50} values of compound 24 with naphthyl substitution at the SH group than those of their corresponding members 15_a , 15_b , 16, 17_a , 17_b , 18 and 19 with alkyl substitutions indicated that substitution with aromatic moieties is advantageous.

With regard to the effect of substituting both the N-3 heteroatom and SH group with different alkyl and aralkyl

moieties, it was found that the introduction two benzyl moieties $(20_a \text{ and } 20_b)$ resulted in a remarkable decrease in anti-proliferative activity in comparison with monosubstitutions (15a, 16, 17_a, 17_b and 19).

The incorporation of a six-membered ring $(21_{a-d}, 22$ and 23) was found to improve biologically versus the incorporation of a five-membered ring (12, 13). Moreover, introduction of substituted aromatic or branched aliphatic moieties at the formed six-membered ring resulted in an increase in anti-proliferative activity.

 Table 2 Docking binding free energies against TS

Comp. ΔG kcal/mol		No. of hydrogen bonds	No. of cationic– π interactions		
12	-48.61	1 (Asn112)	1 (Phe225)		
13	-51.99	3 (His256, Ser216 and Asn226)	2 (Trp109 and Phe225)		
14	-48.61	4 (Asn226, Asp218, Tyr258 and Arg50)	2 (Phe225 and Tyr258)		
15 _a	-45.85	1 (Tyr135)	2 (Trp109 and Tyr258)		
15 _b	-45.20	1 (Tyr135)	2 (Trp109 and Tyr258)		
16	-44.85	3 (Asn226, Ser216 and Arg215)	2 (Phe225 and Tyr258)		
17 _a	-58.16	4 (Gln214, Asn226, His256 and Ser216)	3 (Tyr258, Phe225 and Trp109)		
17 _b	-63.60	3 (Gln214, Asn226 and Ser216)	3 (Ile108 and Met311, Tyr258 and Phe225)		
18	-38.26	1 (Lys308)	1 (Phe225)		
19	-45.85	3 (Ser216, His256 and Arg50)	3 (Phe225, Tyr258 and Trp109)		
20 _a	-40.20	0	2 (Phe225 and Tyr258)		
20 _b	-37.24	0	2 (Phe225 and Tyr258)		
21 _a	-57.22	2 (Ser216 and Asp218)	3 (Tyr258, Trp109 and Phe225)		
21 _b	-50.22	0	3 (Tyr258, Trp109 and Phe225)		
21 _c	-62.40	2 (Ser216 and Asp218)	3 (Tyr258, Trp109 and Phe225)		
21 _d	-63.06	2 (Ser216 and Asp218)	3 (Tyr258, Trp109 and Phe225)		
22	-52.22	0	3 (Tyr258, Trp109 and Phe225)		
23	-58.92	2 (Ser216 and Asp218)	2 (Trp109 and Tyr258)		
24	-83.71	4 (Glu87, Asp218, Gln214 and Asn226)	6 (Phe225, Trp109, Tyr258, Met311, Ile108 and Leu221)		
5-FU	-60.20	3 (Asn226 and Ser216)	2 (Phe225 and Leu221)		
LYA	-90.32	3 (Ala312, Asn112 and Asp218)	6 (Met311, Ile108, Trp109, Phe225, Leu221 and Ile108)		

Fig. 6 Predicted binding mode of compound 24 at the binding site of TS. H-bonds (*yellow dotted lines*), hydrogen atom (*gray*), nitrogen atom (*blue*), oxygen atom (*red*) and sulfur atom (*yellow*). (Color figure online)



Conclusion

A series of 6-aryl-5-cyano thiouracil-based derivatives was designed, synthesized and evaluated for their in vitro antiproliferative activities against HePG-2, MCF-7, HCT-116 and PC-3 cell lines as well as their TS inhibitory activities. Compounds 21_c , 21_d and 24 were the most potent anti-

Fig. 7 Predicted binding mode for compound 17_b at the binding site of TS. Carbonyl group is surrounded by unfavorable interactions with Ile108 and Met311



Fig. 8 3D predicted binding mode analysis for compounds 24 (*left*) and 21_d (*right*) at the TS binding site

for the TS activity with IC₅₀ values between 1.57 ± 0.08 and $5.13 \pm 0.11 \,\mu$ M. Docking studies also confirmed the results obtained from the in vitro thymidylate synthase (TS) assay and anti-proliferation test.

Materials and methods

Chemistry

General

Melting points were measured using a Gallenkamp melting point apparatus and were uncorrected. A Pye-Unicam SP-3-300 infrared spectrophotometer was used to record infrared (IR) spectra using KBr disks, and the data were expressed in wave number (cm⁻¹). ¹H NMR spectra were carried out at 400 and 300 MHz on BrukerAvance III and a Varian Mercury VX-300 NMR spectrometers, respectively; ¹³C NMR spectra were run at 100 and 75 MHz. TMS was used as internal standard in deuterated dimethyl sulfoxide (DMSO-*d*6). Chemical shifts (δ) are expressed in ppm. A Shimadzu GCMS-QP-1000EX mass spectrometer was used to record mass spectra at 70 eV. A CHNO Rapid Analyzer was used to carry out elemental analyses (all compounds were within ±0.4 of the theoretical values). TLC sheets coated with UV fluorescent silica gel Merck 60 F254 plates were used to carry out thin-layer chromatography to monitor the reactions. Compounds **11**_{a,b} are previously reported [33].

General procedure for synthesis of compounds (12–14)

A mixture of compounds $11_{a,b}$ (0.01 mol), anhydrous K₂CO₃ (2.7 g, 0.02 mol) and tetrabutylammonium bromide (TBAB) (0.003 mol) in dry THF (50 mL) was stirred for 30 min at 80–82 °C. Then, appropriate dihalides, namely 2-dibromoethane, acetyl chloride and methylene chloride (0.012 mol), were added. The reaction mixture was stirred vigorously at room temperature for 8 h and monitoring by TLC. At end of the reaction, the reaction mixture was filtered, the filtrate evaporated, and the resulting residue washed three times using excessive amount of petroleum ether and then crystallized to produce the corresponding target compounds **12–14**, respectively.

7-(4-Methoxyphenyl)-5-oxo-2,3-dihydro-5H-thiazolo[3,2

-a]pyrimidine-6-carbonitrile (12) White crystals (yield 77%); m.p. 245–247 °C (EtOH); IR (ν cm⁻¹) 3074 (CH-aromatic), 2999, 2971 (CH-aliphatic), 2212 (CN), 1663 (C=O);¹H NMR (300MHz, DMSO-*d*6) δ (ppm): 3.62 (t, 2H, SCH₂, J = 8 Hz), 3.80 (s, 3H, OCH₃), 4.45 (t, 2H, NCH₂, J =8 Hz), 7.10 (d, 2H, J = 8.8 Hz, H-3 and H-5 of phenyl), 7.90 (d, 2H, J = 9.8 Hz,Ar-H, H-2 and H-6 of phenyl);¹³C NMR (75 MHz, DMSO-*d*6): 28.88, 49.91, 53.44, 92.83, 113.83, 115.80, 116.72, 126.52, 126.93, 129.52, 159.14, 159.63, 165.73, 172.46; MS (m/z): 287 (M⁺ + 2, 10%), 255 (30%) 225 (100%); Anal. Calcd for C₁₄H₁₁N₃O₂S (285.32): C, 58.94; H, 3.89; N, 14.73; Found: C, 58.69; H, 3.68; N, 14.63%.

2,5-Dioxo-7-phenyl-2,3-dihydro-5H-thiazolo[3,2-a]pyrimi dine-6-carbonitrile (13) White crystals (yield 72%); m.p. 238–240 °C (EtOH); IR (ν cm⁻¹) br 3308 (OH), 3028 (CHaromatic), 2998, 2941 (CH-aliphatic), 2227 (CN), 1716 (CO), 1652 (C=O amide); ¹H NMR (300 MHz, DMSOd6) δ (ppm): 4.02 (s, 2H, CH₂), 7.64 (t, 2H, J = 6.9 Hz, Ar-H, H-3 and H-5 of phenyl), 7.94 (d, 2H, J = 6.9 Hz, Ar-H, H-2 and H-6 of phenyl); ¹³C NMR (75 MHz, DMSOd6): 93.43, 116.23, 129.76, 129.76, 129.00, 129.24, 129.24, 132.34, 135.49, 161.49, 165.89, 167.49, 169.64; MS (m/z) 271 (M⁺ +2, 5%), 269 (M⁺, 34%), 229 (47%), 104 (100%), 77 (62%), 51 (38%); Anal. Calcd for C₁₃H₇N₃O₂S (269.00): C, 57.99; H, 2.62; N, 15.61; Found: C, 57.68; H, 2.49; N, 15.51%.

2,2'-[Methylenebis(sulfanediyl)]bis[4-(4-methoxyphenyl)

-6-oxo-1,6-dihydro pyrimidine-5-carbonitrile] (14) White powder (yield 78%); m.p. 293–295 °C (EtOH); IR (ν cm⁻¹) br 3161 (NH), 3107 (CH-aromatic), 2934, 2892 (CHaliphatic), 2225 (CN), 1705 (C=O); ¹H NMR (400 MHz, DMSO-*d*6) δ (ppm): 2.73 (s, 1H, SCH), 2.88 (s, 1H, SCH), 3.80 (s, 3H, OCH₃), 7.11 (d, 4H, J = 8.8 Hz, Ar-H, H-3 and H-5 of phenyl), 7.65 (d, 4H, J = 8.8 Hz, H-2 and H-6 of phenyl), 13.07 (s, 2H, 2NH); ¹³C NMR (100 MHz, DMSO-*d*6): 18.42, 55.33 (2), 95.42 (2), 115.91 (4), 120.96 (2), 130.34 (4), 133.53 (2), 158.57 (2), 163.48 (2), 167.81 (2), 170.14 (2); MS (m/z): 389 (2.5%), 273 (3%) 259 (64%) 134 (100%), 91 (54%); Anal. Calcd for C₂₅H₁₈N₆O₄S₂ (530.58): C, 56.59; H, 3.42; N, 15.84; Found: C, 56.23; H, 3.30; N, 15.73%.

General procedure for synthesis of compounds (15-20)

A mixture of compounds $11_{a,b}$ (0.01 mol), anhydrous K₂CO₃ (2.7g, 0.02 mol) and tetrabutylammonium bromide (TBAB) (0.003 mol) in dry THF (50 mL) was stirred for 30 min at 80–82 °C. Then, appropriate alkyl halide derivative, namely allyl bromide, ethyl chloroacetate, chloroacetic acid, diethyl bromomalonate, ethyl chloroformate and benzyl chloride (0.012 mol), was added. The reaction mixture was stirred vigorously at room temperature for 4 h and monitored by TLC. After completion of the reaction, the reaction mixture was filtered, the filtrate evaporated, and the resulting residue washed three times using excessive amount of petroleum ether and then crystallized to produce the corresponding target compounds **15–20**, respectively.

2-(Allylthio)-4-(4-methoxyphenyl)-6-oxo-1,6-dihydropyri midine-5-carbonitrile (15_a) Yellowish crystals (yield 81%); m.p. 251–253 °C (EtOH); IR(ν cm⁻¹) 3200 (NH), 3072 (CH-aromatic), 2934, 2838 (CH-aliphatic), 2216 (CN), 1651 (C=O)amid; ¹H NMR (400 MHz, DMSO-d6) δ (ppm): 3.85 (s, 3H, OCH₃), 3.94 (d, 2H, SCH₂, J = 6.8 Hz), 5.15 (s, 2H, CH₂ = CH, J = 5.6Hz), 5.95 (q, 1H, CH=, J = 7Hz), 7.13 (d, 2H, J = 7.2Hz, Ar-H, Ar-H, H-3 and H-5 of phenyl), 8.02 (d, 2H,, J = 7.2Hz, Ar-H, H-2 and H-6 of phenyl), 13.65 (s, 1H, NH, D₂O exchangeable); ¹³C NMR (75 MHz, DMSOd6): 13.21, 20.32, 55.44, 61.35, 91.24, 114.27, 114.5, 127.53, 127.92, 132.44, 159.45, 162.44, 168.93, 169.90, 170.68; MS (m/z): 331 (M⁺ + 2, 35%), 299 (M⁺ + 2, 15%), 258 (100%); Anal. Calcd for C₁₅H₁₃N₃O₂S (299.35): C, 60.19; H, 4.38; N, 14.04; Found: C, 59.89; H, 4.18; N, 13.98%.

2-(Allylthio)-6-oxo-4-phenyl-1,6-dihydropyrimidine-5-car bonitrile (15_b) Yellowish crystals (yield 83%); m.p. 238– 240 °C (EtOH); IR(ν cm⁻¹) 3435 (NH), 3018 (CH-aromatic), 2937, 2856 (CH-aliphatic), 2219 (CN), 1654 (C=O); ¹H NMR (400 MHz, DMSO-d6) δ (ppm): 3.90 (d, 2H, SCH₂, J = 6.8 Hz), 5.31 (d, 2H, CH₂ = CH, J = 5.6 Hz), 5.92 (q, 1H, CH=, J = 7 Hz), 7.45 (d.d, 1H, J = 8.1 Hz, Ar-H, H-4 of phenyl), 7.58 (t, 2H, J = 7.5 Hz, Ar-H, H-2 and H-5 of phenyl), 7.94 (d, 2H, J = 7.5 Hz, Ar-H, H-2 and H-6 of phenyl); ¹³C NMR (100 MHz, DMSO-d6): 14.33, 30.24, 62.34, 95.36, 115.13, 126.32, 126.91, 128.21, 128.18, 128.87, 139.25, 161.92, 167.44, 170.45; MS (m/z): 271 (M⁺ + 2, 45%), 269 (M⁺, 28%), 228 (100 %); Anal. Calcd for C₁₄H₁₁N₃OS (269.32): C, 62.44; H, 4.12; N, 15.60; Found: C, 62.15; H, 4.0; N, 15.49%.

2-{[5-Cyano-4-(4-methoxyphenyl)-6-oxo-1,6-dihydropyri midin-2-yl]thio} acetic acid (16) White powder (yield 75%); m.p. 233–235 °C (EtOH); IR (ν cm⁻¹) 2750–3514 (br OHcarboxylic), 3200 (NH), 3009 (CH-aromatic), 2933, 2842 (CH-aliphatic), 2219 (CN), 1717 (C=O carboxylic) acid, 1651 (C=O); ¹H NMR (400 MHz, DMSO-*d*6) δ (ppm): 3.86 (s, 3H, OCH₃), 4.06 (s, 2H, SCH₂), 7.10 (d, 2H, J = 8 Hz, Ar-H, H-3 and H-5 of phenyl), 8.01 (d, 2H, J = 8 Hz, Ar-H, H-2 and H-6 of phenyl), 13.17 (s, 1H, NH, D₂O exchangeable), 13.23 (s, 1H, OH, D₂O exchangeable); ¹³C NMR (100 MHz, DMSO-*d*6): 18.92, 33.52, 56.01, 56.42, 91.74, 114.45, 116.73, 127.52, 131.41, 161.58, 162.71, 165.28, 166.58, 169.67; MS (m/z) 317 (M⁺, 3.1 %), 298 (100%); Anal. Calcd for C₁₄H₁₁N₃O₄S (317.29): C, 52.99; H, 3.49; N, 13.24; Found: C, 52.79; H, 3.31; N, 13.15%.

Ethyl 2-((5-cyano-4-(4-methoxyphenyl)-6-oxo-1,6-dihydro *pyrimidin-2-yl) thio)acetate* (*17_a*) Gray powder (yield 78%); m.p. 212–214 °C (EtOH); IR (ν cm⁻¹) 3301 (NH), 3013 (CH-aromatic), 2982 (CH-aliphatic), 2218 (CN), 1737 (CO ester), 1655 (C=O amide); ¹H NMR (400 MHz, DMSOd6) δ (ppm): 1.23 (t, 3H, CH₃, J = 7.2 Hz), 3.86 (s, 2H, SCH_2), 4.01 (s, 3H, OCH₃), 4.07 (q, 2H, OCH₂, J =6Hz), 7.16 (d, 2H, J = 8.7 Hz, Ar-H, H-3 and H-5 of phenyl), 8.04 (d, 2H, J = 8.7 Hz, Ar-H, H-2 and H-6 of phenyl) 13.8 (s, 1H, NH, D₂O exchangeable); ¹³C NMR (100 MHz, DMSO-d6): 14.38, 33.35, 56.03, 61.77, 91.87, 114.43, 115.36, 116.63, 127.42, 131.31, 133.70, 162.81, 164.93, 166.4, 168.48, 169.69; MS (m/z): 347 $(M^+ + 2)$, 18%), 300 (33%), 258 (75%), 122 (100%); Anal. Calcd for C₁₆H₁₅N₃O₄S: (345.37): C, 55.64; H, 4.38; N, 12.17; Found: C, 55.37; H, 4.19; N, 12.05%.

Ethyl2-[(5-cyano-6-oxo-4-phenyl-1,6-dihydropyrimidin-2

-yl]thio)acetate (17_b) White crystal (yield 75%); m.p. 240–242 °C (EtOH); IR(ν cm⁻¹) 3203 (NH), 3013 (CH-aromatic), 2985, 2941 (CH-aliphatic), 2222 (CN), 1735 (C=O ester), 1660 (CO amide); ¹H NMR (400 MHz, DMSO-*d*6) δ (ppm): 1.09 (t, 3H, CH₃, J = 7Hz), 4.11 (s, 2H, SCH₂), 4.05 (q, 2H, OCH₂, J = 7.2 Hz), 7.58 (dd, 1H, J = 6.9Hz, Ar-H, H-4 of phenyl), 7.74 (t, 2H, J = 7.8Hz, Ar-H, H-3 and H-5 of phenyl), 7.90 (d, 2H, J = 7.8Hz, Ar-H, H-2 and H-6 of phenyl); ¹³C NMR (100 MHz, DMSO-*d*6): 14.80, 15.92, 59.83, 94.32, 116.31, 126.22, 126.86, 128.15, 128.38, 128.79, 137.39, 159.93, 166.36, 169.35, 170.88; MS (m/z): 315 [M⁺, 12 %), 269 (11%), 242 (27%), 104 (26%), 91(100%); Anal. Calcd for C₁₅H₁₃N₃O₃S: (315.35): C, 57.13; H, 4.16; N, 13.33; Found: C, 56.95; H, 3.99; N, 13.21%.

Diethyl2-{[5-cyano-4-(4-methoxyphenyl)-6-oxo-1,6-dihy

dropyrimidin-2-l] thio}vmalonate (18) White crystal (yield 77%); m.p. 214–216°C (EtOH IR (ν cm⁻¹) 3205 (NH), 3055 (CH-aromatic), 2992, 2942 (CH-aliphatic), 2226 (CN),

1705 (C=O ester), 1645 (C=O amide); ¹H NMR (400 MHz, DMSO-*d*6) δ (ppm): 1.20 (t, 6H, 2CH₃, J = 7.2 Hz), 3.88 (s, 3H, OCH₃), 4.02 (q, 4H, 2CH₂, J = 7.2 Hz), 5.01 (s, 1H, SCH), 7.25 (d, 2H, J = 9.6 Hz, Ar-H, H-3 and H-5 of phenyl), 7.96 (d, 2H, J = 8.8 Hz, Ar-H, H-2 and H-6 of phenyl), 12.50 (s, 1H, NH, D₂O exchangeable); ¹³C NMR (100 MHz, DMSO – *d*6): 14.32, 14.24, 45.33, 55.32, 93.17, 114.64, 114.78, 127.46 (2), 127.94 (2), 160.51 (2), 160.96 (2), 164.35, 174.37, 166.76, 170.85; MS (*m*/*z*): 419 (M⁺ + 2, 77%), 259 (100%),; Anal. Calcd for C₁₉H₁₉N₃O₆S: (417.44): C, 54.67; H, 4.59; N, 10.07; Found; C, 54.39; H, 4.45; N, 9.97%.

S-[5-Cyano-4-(4-methoxyphenyl)-6-oxo-1,6-dihydropyri

midin-2-yl]propane thioate (19) Yellow crystal (yield 79%); m.p. 216–218 °C (EtOH); IR (ν cm⁻¹) 3151 (NH), 3078 (CH-aromatic), 2996, 2939 (CH-aliphatic), 2239 (CN), 1785 (C=O ester), 1701 (C=O amide); ¹H NMR (400 MHz, DMSO-*d*6) δ (ppm): 1.04 (t, 3H, CH₃, J = 7.2 Hz), 3.85 (s, 3H, OCH₃), 4.23 (q, 2H, CH₂, J = 7.2 Hz), 7.25 (d, 2H, J = 8.8 Hz, Ar-H, H-3 and H-5 of phenyl), 7.96 (d, 2H, Ar-H, J = 8.8 Hz, Ar-H, H-2 and H-6 of phenyl), 12.50 (s, 1H, NH, D₂O exchangeable); ¹³C NMR (100 MHz, DMSO-*d*6): 14.72, 55.67, 62.79, 93.70, 114.68, 114.85, 115.65, 127.45, 127.93, 129.89, 158.25, 159.41, 159.97, 166.67, 170.78; MS (m/z): 389 (M⁺, 2.5%), 273 (3%) 259 (64%) 134 (100%); Anal. Calcd for C₁₅H₁₃N₃O₃S; (331.06): C, 54.37; H, 3.95; N, 12.68; Found C, 54.08; H, 3.83; N, 12.55%.

1-Benzyl-2-(benzylthio)-4-(4-methoxyphenyl)-6-oxo-1,6-

dihydropyrimidine-5-carbonitrile (20*a*) Reddish crystals (yield 65%); m.p. 170–172 °C (EtOH); IR (ν cm⁻¹) 3060 (CH-aromatic), 2964, (CH-aliphatic), 2212 (CN), 1674 (C=O); 1H NMR (400 MHz, DMSO-*d*6) δ (ppm): 3.84 (s, 3H, CH₃O), 4.50, 4.59 (s, s, 2H, SCH₂), 5.2 and 5.6 (s, s, 2H, NCH₂), 7.13–8.05 (m, 14H, Ar-H); ¹³C NMR (100 MHz, DMSO-*d*6): 35.18, 36.78, 47.96, 56.00, 69.93, 87.32, 91.42, 114.60, 114.65, 115.44, 116.71, 127.73, 128.51, 128.98, 129.56, 131.44, 134.57, 135.73, 137.63, 160.75, 162.71, 164.75, 166.14, 167.81, 169.55, 173.97; MS (*m*/*z*): 439 (M⁺, 100 %), 330 (10%), 270 (12%), 211 (40%), 139 (62%), 56 (51%) . Anal. Calcd for C₂₆H₂₁N₃O₂S; (439.53): C, 71.05; H, 4.82; N, 9.56; Found: C, 70.75; H, 4.76; N, 9.45%.

2-(Benzylthio)-6-oxo-1,4-diphenyl-1,6-dihydropyrimidine-5-carbonitrile (**20**_b) White crystals (Yield 69%); m.p. 185–187 °C (EtOH); IR(ν cm⁻¹) 3078 (CH-aromatic), 2996, 2939(CH-aliphatic), 2212 (CN), 1674 (C=O); ¹HNMR (400 MHz, DMSO-d6) δ (ppm): 4.50 (s, 2H, NCH₂), 5.28 (s, 2H, SCH₂), 7.31–7.98 (m,15H, Ar-H); ¹³C NMR (100 MHz, DMSO-d6): 137.64, 45.96, 94.23, 116.43, 126.46(2), 126.68, 126.93 (2), 127.21 (2), 127.61, 127.72 127.93 (2), 128.44 (2), 128.65, 128.92 (2), 136.44, 136.94, 160.28, 162.83, 170.58; MS (*m*/*z*): 409 (M⁺, 1.17%), 318 (13 %), 148 (19%), 141 (33%), 139 (100%), 91 (53%); Anal. Calcd for $C_{25}H_{19}N_3OS$; (409.51): C, 73.33; H, 4.68; N, 10.26; Found: C, 73.06; H, 4.51; N, 10.16%.

General procedures for synthesis of compounds (21-24)

A mixture of compounds $11_{a,b}$ (1.0 mmol), appropriate primary amines, namely *p*-anisidine, *p*-toluidine, 2-aminopyrimidine, iso-butylamine and β -naphthylamine (1.1 mmol), and formaldehyde (2 mL) was stirred in dioxane (40 mL) at room temperature for 3 h. The resulting precipitate was collected by filtration, washed with water several times and dried. The crude product was crystallized using a proper solvent to give the corresponding title compounds **21–24**, respectively.

3,8-Bis(4-methoxyphenyl)-6-oxo-3,4-dihvdro-2H,6H-pyri mido[2,1-b][1,3,5] thiadiaine-7-carbonitrile (21a) Pale yellow crystal (yield 91%); m.p. 225-228°C (EtOH); IR (v cm⁻¹) 3055 (CH-aromatic), 2994 (CH-aliphatic), 2214 (CN) 1645 (C=O); ¹H NMR (400 MHz, DMSO-d6) δ (ppm): 3.69(s, 3H, OCH₃), 3.83 (s, 3H, OCH₃), 5.46 (s, 2H, S-CH2-N), 5.67 (s, 2H, N-CH2-N), 6.92 (d, 2H, Ar-H, H-3 and H-5 of phenyl, J = 8 Hz), 7.08 (d, 2H, Ar-H, H-2 and H-6 of phenyl, J = 8 Hz), 7.10 (d, 2H, Ar-H, H-3 and H-5 of phenyl, J = 9.2 Hz), 7.90 (d, 2H, Ar-H, H-2 and H-6 of phenyl, J = 9.2 Hz); ¹³C NMR (100 MHz, DMSO-*d*6): 55.23, 55.86, 70.27, 93.29, 114.26 (2), 115.24(2), 115.93 (2), 127.72 (2), 129.23, 142.25, 153.24, 159.28, 160.98, 163.29, 170.84; MS (m/z): 408 $(M^+ + 2, 100\%)$; Anal. Calcd for C₂₁H₁₈N₄O₃S; (406.46): C, 62.06; H, 4.46; N, 13.78; Found: C, 61.89; H, 4.29; N, 13.69%.

8-(4-Methoxyphenyl)-6-oxo-3-(p-tolyl)-3,4-dihydro-2H, 6H-pyrimido[2,1-b] [1,3,5] thiadiaine-7-carbonitrile (21_b)

Yellowish powder (yield 88%); m.p. 228–230 °C (EtOH); IR (ν cm⁻¹) 3102 (CH-aromatic), 2915 (CH-aliphatic), 2219 (CN) 1658 (C=O); ¹H NMR (400 MHz, DMSO-*d*6) δ (ppm): 2.21 (s, 3H, CH₃), 3.83 (s, 3H,-OCH₃), 5.49 (s, 2H, S–CH₂–N), 5.71 (s, 2H, N–CH₂-N), 7.04 (d, 2H, Ar-H, H-2 and H-6 of *P* – tolyl, *J* = 6.8 Hz), 7.07 (d, 2H, Ar-H, H-3 and H-5 of *P*-methoxy phenyl, *J* = 8.4 Hz), 7.15 (d, 2H, Ar-H,H-3 and H-5 of *P*-tolyl, *J* = 6.8 Hz), 7.90 (d, 2H, Ar-H, H-2 and H-6 of *P* –methoxy phenyl, *J* = 8.4 Hz); ¹³C NMR (100 MHz, DMSO-*d*6): 21.56, 55.64, 55.96, 70.52, 93.59, 112.44 (2), 114.33 (2), 115.41, 127.46 (2),129.15, 129.73(2), 130.73, 160.44, 162.42, 163.64, 170.75; MS (*m*/*z*): 392 (M⁺ + 2, 24%), 225 (100%); Anal. Calcd for C₂₁H₁₈N₄O₂S; (390.46): C, 64.60; H, 4.65; N, 14.35; Found: C, 64.39; H, 4.46; N, 14.29%.

3-(4-Methoxyphenyl)-6-oxo-8-phenyl-3,4-dihydro-2H,6H -pyrimido[2,1-b] [1,3,5] thiadiazine-7-carbonitrile (21c) Pale yellow crystals (yield 79%); m.p. 225–227 °C (EtOH / Benzene); $IR(\nu \text{ cm}^{-1})$ 3041(CH-aromatic), 2955 (CH- aliphatic), 2215 (CN), 1652 (C=O); ¹H NMR (300 MHz, DMSO-*d*6) δ : (ppm): 3.67 (s, 3H, OCH₃), 5.45 (s, 2H, S–CH₂–N), 5.76 (s, 2H, N–CH₂–N), 6.90 (d, 2H, Ar-H, H-3 and H-3 of *P*-methoxy phenyl, J = 8.2 Hz), 7.10 (d, 2H, Ar-H, H-2 and H-6 of *P*-methoxy phenyl, J = 8.2 Hz), 7.33 (dd, 1H, J = 7.00 Hz, Ar-H, H-4 of phenyl), 7.53 (t, 2H, J = 8.1 Hz, Ar-H, H-3 and H-5 of phenyl), 7.82 (d, 2H, J = 8.1 Hz, Ar-H, H-2 and H-6 of phenyl); ¹³C NMR (75 MHz, DMSO-*d*6): 55.00, 55.66, 62.21, 92.30, 115.35, 115.99, 118.69, 129.01, 129.03, 132.28, 132.30, 134.97 134.99, 137.33, 137.34 137.35, 155.32, 153.63, 165.45, 166.27; MS (m/z): 378 (M⁺ + 2, 55%), 229 (100%); Anal. Calcd for C₂₀H₁₆N₄O₂S: (376.43): C, 63.81; H, 4.28; N, 14.88; Found: C, 63.54; H, 4.17; N, 14.79%.

6-Oxo-8-phenyl-3-(p-tolyl)-3,4-dihydro-2H,6H-pyrimido

[2,1-b][1,3,5]thiadiazine-7-carbonitrile (21_d) White crystals (yield 85%); m.p. 220-222°C (EtOH, Benzene); IR (v cm⁻¹) 3049 (CH-aromatic), 2915 (CH-aliphatic), 2214 (CN), 1653 (C=O); ¹H NMR (300 MHz, DMSO-*d*6) δ: (ppm): 2.25 (s, 3H, CH₃), 5.50 (s, 2H, S-CH₂-N), 5.73 (s, 2H, N-CH₂-N), 7.65 (d, 2H, Ar-H, H-2 and H-6 of *p*-tolyl), 7.16 (d, 2H, Ar-H, H-3 and H-5 of *p*-tolyl), 7.50 (dd, 1H, J = 7.4 Hz, Ar-H, H-4 of phenyl), 7.66 (t, 2H, J = 7.4 Hz, Ar-H, H-3 and H-5 of phenyl), 7.83 (d, 2H, J = 8.4 Hz, Ar-H, H-2 and H-6 of phenyl); ¹³C NMR (75 MHz, DMSOd6): 22.14, 55.46, 70.30, 94.20, 112.35 (2), 115.75, 126.37 (2), 127.14, 128.30(2), 129.26 (2), 130.66, 137.72, 147.87, 160.19, 163.75, 170.78; MS (m/z): 362 $(M^+ + 2, 33\%)$, 229 (100%); Anal. Calcd for C₂₀H₁₆N₄OS: (360.44) C, 66.65; H, 4.47; N, 15.54; Found: C, 66.37; H, 4.53; N, 4.41%.

6-Oxo-8-phenyl-3-(pyrimidin-2-yl)-3,4-dihydro-2H,6H-py rimido[2,1-b] [1,3, 5] thiadiazine-7-carbonitrile (22) White crystals (yield 85%); m.p. 248–250 °C (EtOH); IR (ν cm⁻¹) 3059 (CH-aromatic), 2940 (CH-aliphatic), 2230 (CN), 1679 (C=O); ¹H NMR (300 MHz, DMSO-d6) δ: (ppm): 5.52 (s, 2H, S–CH₂–N), 5.77 (s, 2H, N–CH₂–N), 6.87–8.55 (m, 8H, Ar-H); MS (m/z): 348 (M⁺, 45%), 229 (48%), 108 (100%); ¹³C NMR (75 MHz, DMSO-d6): 55.64, 70.34, 94.25, 115.66, 115.89, 126.63(2), 127.55, 128.94(2), 135.34, 158.24(2), 160.34, 162.35, 162.96, 171.19; MS (m/z): 350 (M⁺ + 2, 100%), 228 (22%) Anal. Calcd for C₁₇H₁₂N₆OS (348.38) C, 58.61; H, 3.47; N, 24.12; Found: C, 58.29; H, 3.29; N, 24.04%.

3-Isobutyl-6-oxo-8-phenyl-3,4-dihydro-2H,6H-pyrimido

[2,1-b][1,3,5]thiadiazine-7-carbonitrile (23) Pale yellow crystal (yield 82%); m.p. 218–220 °C (EtOH); IR (ν cm⁻¹) 3150 (CH-aromatic), 2954 (CH-aliphatic), 2215 (CN), 1670 (C=O); ¹H NMR (300 MHz, DMSO-*d*6) δ (ppm): 0.92 (d, 6H, 2CH₃, J = 5.7Hz), 1.83 (m, H, CH, J = 7.2Hz),

1.83 (d, 2H, CH₂, J = 7.2 Hz), 4.90 (s, 2H, S–CH₂–N), 5.09 (s, 2H, N–CH₂–N), 7.51 (dd, 1H, J = 7.49 Hz, Ar-H, H-4 of phenyl), 7.56 (t, 2H, J = 8.2 Hz, Ar-H, H-3 and H-5 of phenyl), 7.71 (d, 2H, J = 8.2 Hz, Ar-H, H-2 and H-6 of phenyl); ¹³C NMR (75 MHz, DMSO-*d*6): 21.32(2), 25.38, 54.34, 62.45, 68.24, 93.93, 115.45, 126.75(2), 127.83, 128.94(2), 136.65, 160.35, 162.86, 170.70; MS (m/z): 326 (M⁺, 95%), 293 (28%); Anal. Calcd for C₁₇H₁₈N₄OS; (326.42): C, 62.55; H, 5.56; N, 17.16; Found: C, 62.23; H, 5.45; N, 17.08%.

2-(((Naphthalen-2-ylamino)methyl)thio)-6-oxo-4-phenyl-

I,6-dihydropyrimidine-5-carbonitrile (*24*) Reddish brown crystal (yield 77%), m.p. 165–167 °C (EtOH); IR(ν cm⁻¹) 3163 (NH), 3031 (CH-aromatic), 2852 (CH-aliphatic), 2221 (CN), 1671 (C=O); ¹H NMR (300 MHz, DMSO-*d*6) δ (ppm): 3.51 (s, 2H, CH₂), 7.21 (s, 1H, NH-naphthyl, D₂O exchangeable), 7.55–7.81 (m, 11H, Ar-H), 12.99 (s, 1H, NH pyrimidine, D₂O exchangeable); ¹³C NMR (75 MHz, DMSO-*d*6): 45.83, 94.41, 105.52, 115.32, 118.64, 121.43, 121.83, 124.42, 125.44 (2), 126.46, 126.84, 127.14, 127.96 (2), 128.48, 136.46, 138.94, 144.83, 160.46, 167.35, 171.95; MS (*m*/*z*): 386 (M⁺ + 2, 17%), 288 (100%); Anal. Calcd for C₂₂H₁₆N₄OS (384.46): C, 68.73; H, 4.20; N, 14.57; Found: C, 68.51; H, 4.10; N, 14.44%.

Biological evaluation

In vitro anti-proliferative activity

Four human cancer cell lines, namely hepatocellular carcinoma (HePG-2), mammary gland (MCF-7), colorectal carcinoma (HCT-116) and human prostate cancer (PC-3), are used to determine in vitro the anti-proliferation properties of the synthesized compounds. The tested cell lines were supplied from the US National Cancer Institute. The reported standard procedure described by Skehan et al. [32] was utilized in this test as follows.

The tested cells were plated in 96-well microplates; the total volume per well was adjusted at $100 \,\mu$ L. Then, incubation of cells was performed at $37 \,^{\circ}$ C, $5\% \,$ CO₂, $95\% \,$ air and 100% relative humidity for 24h before addition of synthesized compounds. After 24h, only two plates of each cell line were selected and fixed in situ with TCA, in order to exemplify a measurement of the cell population for each cell line during drug application. The title compounds and fluorouracil, the reference drug, were dissolved in DMSO at 400-fold the desired final maximum test concentration and stored at freezing point prior to use. During addition of drug, the frozen concentrate was dissolved and diluted to twice the desired final maximum test concentration, the desired final drug concentration, different tested compound dilutions (100 mL) were

added to the appropriate microtiter wells containing 100 mL of medium. The tested compounds as well as 5-fluorouracil as reference drug were added. Then, the plates were incubated for an additional 48h at 37°C, 5% CO₂, 95% air and 100% relative humidity. The assay was terminated by addition of cold TCA for adherent cells followed by incubation for 60 min at 4 ° C. The supernatant was removed, and the plates were washed five times with excessive water and dried. A solution of 0.4% (w/v) sulforhodamine B (100 mL) in 1% acetic acid was added to each well, followed by incubation at room temperature for 10 min. After staining, the plates were washed with 1% acetic acid to remove unbound dye and air-dried. Bound stain was dissolved with 10 mM Trizma base. Spectrophotometric assay of the optical density (OD) of each well was determined at 564 nm with an ELISA microplate reader (ChroMate-4300, FL, USA). Boltzmann sigmoidal concentration response curve was used to calculate the IC₅₀ values through the nonlinear regression fitting models (GraphPad Prism, version 5). The means of three separate experiments was reported as final result. ANOVA test was used to analyze the statistical differences, wherein the differences were considered to be significant at p < 0.05.

In vitro assay of thymidylate synthase (TS) activity

The kinase activity of TS was measured spectrophotometrically at pH 7.4 and 30°C in a mixture containing 0.1 M 2-mercaptoethanol, 0.0003 M (6R,S)-tetrahydrofolate, 0.02 M MgCl₂, 0.012 M formaldehyde, 0.04 M TrisHCl, 0.001 M dUMP and 0.00075 M Na EDTA according to the reported procedure described by Wahba and Friedkin [37]. In general, to initiate the reaction, the enzyme was added in the absence of inhibitor producing a change in absorbance at 340 nm of 0.016/min. Then, four inhibitor concentrations were used to determine the percent inhibition. Next, concentration-inhibition response curve for the test compounds was generated to determine median inhibitory concentration (IC₅₀). The obtained data were compared with 5-fluorouracil as a standard TS inhibitor. The standard deviations for determination of the 50% points were within $\pm 10\%$ of the values given.

Molecular docking

All docking studies were performed using AutoDock 4.2 program [38,39]. The structure of the synthesized compounds and reference ligands was drawn using ACD/ChemSketch (version 12.0, Toronto, Canada) [40], and their geometry was optimized by Autodock 4.2. The scoring function used is empirically derived for empirical binding free energy force field that allows the prediction of binding free energies for docked ligands. The 3D crystal structure of the thymidylate synthase (TS) was downloaded from the Brookhaven Protein Databank (PDB ID: 1JU6, 2.89 Å resolution) [41]. All bound waters ligands and cofactors were removed from the protein prior to the docking process. Then, SPDBV-Swiss-pdb viewer [42] was used to minimize the energy. Next, hydrogen atoms were added for the minimized receptor. Kollman charges and solvation parameters were applied. In order to include all the amino acid residues, the precalculated grid maps are set at the size of 60, 60 and 60 Å (x, y and z). The spacing between grid points was 0.575 Å. The Lamarckian genetic algorithm (LGA) was utilized in the docking process. All the test compounds were docked within the prepared grid. The crystallized pose of 5-fluorouracil and pemetrexed (LYA) was used as standards. The output from AutoDock was further analyzed with PyMOL software package [43].

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Compliance with ethical standards

Conflict of interest This work was funded by the authors, and there is no any conflict of interest.

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